

Puri Sharma Pathania Physical Chemistry

Advanced Inorganic Chemistry - Volume II is a concise book on basic concepts of inorganic chemistry. Beginning with Coordination Chemistry, it presents a systematic treatment of all Transition and Inner-Transition chemical elements and their compounds according to the periodic table. Special topics such as Pollution and its adverse effects, chromatography, use of metal ions in biological systems, to name a few, are discussed to provide additional relevant information to the students. It primarily caters to the undergraduate courses (Pass and Honours) offered in Indian universities.

The Revised Edition Retains The Essential Theories Of Nuclear Structure And Stability, Radioactivity And The Principles Of Fission, Fusion And Breeder Reactors Of The Earlier Editions. The Preparation Of The More Commonly Used Radioisotopes And Their Uses As Tracers In Research, Medicine, Agriculture And Industry Are Described. The Book Also Covers The Elements Of Radiation And Radiochemistry Illustrated With Additional Examples. The Section On Mossbauer Effect Is Retained. The Chapter On The Detection And Measurement Of Radioactivity Is Revised To Include Thermo Luminescence And Cerenkov Detectors. New Additions In The Present Edition Include A Whole Chapter On The Separation And Uses Of Stable And Radioactive Isotopes Needed In Bulk Amounts In The Atomic Age. How An Extension Of Basic Principles Of Nuclear Magnetic Resonance (Nmr) Has Led To The Sophisticated Magnetic Resonance Imaging (Mri), The Latest Diagnostic Tool In Medicine Is Discussed Lucidly. Another Chapter Is Added Entitled A Roll-Call Of Elementary Particles, Wherein The Baffling Properties Of Quarks And Gluons, With Their Esoteric Flavours, Colours, Strangeness And Charm Are Reviewed Showing How Their Scientific Characteristics Tend To Merge In Philosophy. The Book Meets The Needs Of Honours And Post-Graduate Students Offering Nuclear, Radiation And Radiochemistry.

Both elementary inorganic reaction chemistry and more advanced inorganic theories are presented in this one textbook, while showing the relationships between the two. This Book Is Especially Designed According To The Model Curriculum Of M.Sc. (Prev.) (Pericyclic Reactions) And M.Sc. (Final) (Photochemistry Compulsory Paper Viii) Suggested By The University Grants Commission, New Delhi. As Far As The Ugc Model Curriculum Is Concerned, Most Of The Indian Universities Have Already Adopted It And The Others Are In The Process Of Adopting The Proposed Curriculum. In The Present Academic Scenario, We Strongly Felt That A Comprehensive Book Covering Modern Topics Like Pericyclic Reactions And Photochemistry Of The Ugc Model Curriculum Was Urgently Needed. This Book Is A Fruitful Outcome Of Our Aforesaid Strong Feeling. Besides M.Sc. Students, This Book Will Also Be Very Useful To Those Students Who Are Preparing For The Net (Csir), Slet, Ias, Pcs And Other Competitive Examinations. The Subject Matter Has Been Presented In A Comprehensive, Lucid And Systematic Manner Which Is Easy To Understand Even By Self Study. The Authors Believe That Learning By Solving Problems Gives More Competence And Confidence In The Subject. Keeping This In View, Sufficiently Large Number Of Varied Problems For Self Assessment Are Given In Each Chapter. Hundred Plus Problems With Solutions In The Last Chapter Is An Important Feature Of This Book.

For B.Sc. Part I, II & III Classes of all Indian Universities and also covering U.G.C.

model curriculum. Authentic, simple, to the point and modern account of each and every topic. Relevant, Clear, well labelled diagrams. Easy to understand treatment of most difficult and intricate topic. Questions from university papers of various Indian Universities

An advanced-level textbook of physical chemistry for the graduate (B.Sc) and postgraduate (M.Sc) students of Indian and foreign universities. This book is a part of four volume series, entitled "A Textbook of Physical Chemistry – Volume I, II, III, IV". CONTENTS: Chapter 1. Quantum Mechanics – I: Postulates of quantum mechanics; Derivation of Schrodinger wave equation; Max-Born interpretation of wave functions; The Heisenberg's uncertainty principle; Quantum mechanical operators and their commutation relations; Hermitian operators (elementary ideas, quantum mechanical operator for linear momentum, angular momentum and energy as Hermitian operator); The average value of the square of Hermitian operators; Commuting operators and uncertainty principle (x & p ; E & t); Schrodinger wave equation for a particle in one dimensional box; Evaluation of average position, average momentum and determination of uncertainty in position and momentum and hence Heisenberg's uncertainty principle; Pictorial representation of the wave equation of a particle in one dimensional box and its influence on the kinetic energy of the particle in each successive quantum level; Lowest energy of the particle. Chapter 2. Thermodynamics – I: Brief resume of first and second Law of thermodynamics; Entropy changes in reversible and irreversible processes; Variation of entropy with temperature, pressure and volume; Entropy concept as a measure of unavailable energy and criteria for the spontaneity of reaction; Free energy, enthalpy functions and their significance, criteria for spontaneity of a process; Partial molar quantities (free energy, volume, heat concept); Gibb's-Duhem equation. Chapter 3. Chemical Dynamics – I: Effect of temperature on reaction rates; Rate law for opposing reactions of 1st order and 2nd order; Rate law for consecutive & parallel reactions of 1st order reactions; Collision theory of reaction rates and its limitations; Steric factor; Activated complex theory; Ionic reactions: single and double sphere models; Influence of solvent and ionic strength; The comparison of collision and activated complex theory. Chapter 4. Electrochemistry – I: Ion-Ion Interactions: The Debye-Huckel theory of ion-ion interactions; Potential and excess charge density as a function of distance from the central ion; Debye Huckel reciprocal length; Ionic cloud and its contribution to the total potential; Debye - Huckel limiting law of activity coefficients and its limitations; Ion-size effect on potential; Ion-size parameter and the theoretical mean-activity coefficient in the case of ionic clouds with finite-sized ions; Debye - Huckel-Onsager treatment for aqueous solutions and its limitations; Debye-Huckel-Onsager theory for non-aqueous solutions; The solvent effect on the mobility at infinite dilution; Equivalent conductivity (?) vs. concentration $c^{1/2}$ as a function of the solvent; Effect of ion association upon conductivity (Debye- Huckel - Bjerrum equation). Chapter 5. Quantum Mechanics – II: Schrodinger wave equation for a particle in a three dimensional box; The concept of degeneracy among energy levels for a particle in three dimensional box; Schrodinger wave equation for a linear harmonic oscillator & its solution by polynomial method; Zero point energy of a particle possessing harmonic motion and its consequence; Schrodinger wave equation for three dimensional Rigid rotator; Energy of rigid rotator; Space quantization; Schrodinger wave equation for hydrogen atom, separation of variable in polar spherical coordinates and its solution; Principle, azimuthal and magnetic quantum numbers and the magnitude of their values; Probability distribution function; Radial distribution function; Shape of atomic orbitals (s,p & d). Chapter 6. Thermodynamics – II: Clausius-Clayperon equation; Law of mass action and its thermodynamic derivation; Third law of thermodynamics (Nernst heat theorem, determination of absolute entropy, unattainability of absolute zero) and its limitation; Phase diagram for two completely miscible components systems; Eutectic systems, Calculation of eutectic point; Systems forming solid compounds $A_x B_y$ with congruent and incongruent melting points; Phase diagram and thermodynamic

treatment of solid solutions. Chapter 7. Chemical Dynamics – II: Chain reactions: hydrogen-bromine reaction, pyrolysis of acetaldehyde, decomposition of ethane; Photochemical reactions (hydrogen - bromine & hydrogen -chlorine reactions); General treatment of chain reactions (ortho-para hydrogen conversion and hydrogen - bromine reactions); Apparent activation energy of chain reactions, Chain length; Rice-Herzfeld mechanism of organic molecules decomposition(acetaldehyde); Branching chain reactions and explosions (H₂-O₂ reaction); Kinetics of (one intermediate) enzymatic reaction : Michaelis-Menton treatment; Evaluation of Michaelis 's constant for enzyme-substrate binding by Lineweaver-Burk plot and Eadie-Hofstae methods; Competitive and non-competitive inhibition. Chapter 8.

Electrochemistry – II: Ion Transport in Solutions: Ionic movement under the influence of an electric field; Mobility of ions; Ionic drift velocity and its relation with current density; Einstein relation between the absolute mobility and diffusion coefficient; The Stokes- Einstein relation; The Nernst -Einstein equation; Walden's rule; The Rate-process approach to ionic migration; The Rate process equation for equivalent conductivity; Total driving force for ionic transport, Nernst - Planck Flux equation; Ionic drift and diffusion potential; the Onsager phenomenological equations; The basic equation for the diffusion; Planck-Henderson equation for the diffusion potential.

The third edition presents thoroughly revised and updated text in a simple and easy-to understand language. The book systematically presents the principles, requirements, methods and interpretation of results of various experiments performed in practical biochemistry classes.

A leading book for 80 years, Silbey's Physical Chemistry features exceptionally clear explanations of the concepts and methods of physical chemistry for students who have had a year of calculus and a year of physics. The basic theory of chemistry is presented from the viewpoint of academic physical chemists, but the many practical applications of physical chemistry are integrated throughout the text. The problems in the text also reflect a skillful blend of theory and practical applications. This text is ideally suited for a standard undergraduate physical chemistry course taken by chemistry, chemical engineering, and biochemistry majors in their junior or senior year.

The book, name Physical Chemistry has been written for the students of B.Sc. at different Universities of India, is mainly for examination oriented text book for those, who wants to achieve good concept and good results in their academic examinations, which makes capable to enroll into the Postgraduation courses also

Essentials of Physical Chemistry is a classic textbook on the subject explaining fundamentals concepts with discussions, illustrations and exercises. With clear explanation, systematic presentation, and scientific accuracy, the book not only helps the students clear misconceptions about the basic concepts but also enhances students' ability to analyse and systematically solve problems. This bestseller is primarily designed for B.Sc. students and would equally be useful for the aspirants of medical and engineering entrance examinations.

Presents a new nomenclature and covers recently discovered systems. Includes a detailed study of conformational analysis of acyclic and alicyclic compounds, the relation between conformation and reactivity, and other aspects of stereochemistry, such as substitution, addition and elimination reactions.

Includes numerous examples and illustrations from the Natural Product Area.

Key Features: Concepts built from strong and ground origins. Brief presentation on atomic, hybrid and molecular orbital concepts. Molecular structure and symmetry presented in pedagogical manner. Illustrations with a variety of molecular examples. Self-study exercises for thorough understanding. crossword

puzzles provide test of learning. Appendices at the end provide an essential supplement. About the Book: This book is designed with an exclusive coverage of symmetry & structure of molecules. Also, the teachers would find a classroom-friendly narration of all the topics presented in the book. An exclusive excursion-like treatment is given for the concepts of structure, symmetry and orbitals (atomic, hybrid and molecular) with a semi-pedagogical coverage. The primary focus of the book is on 'root-learning' than on 'rote-learning', paving the way for strong foundations. Secondly, the treatment given in the book helps in learning the correct concept by both the teacher and the taught. The method of presentation chosen for the book is the one that is well tested in the classroom for few decades. The book attempts a systematic approach in mastering the subject layer by layer and a smooth transition is maintained throughout from chapter to chapter for a successful take off.

Resulting from the premier forum for pesticide development and use, this volume provides comprehensive coverage and even captures emerging technologies within the industry. All facets of pesticides are addressed here, including agriculture, agrochemicals, and environmental health aspects, as well as such global issues as food quality and safety.

Principles of Physical Chemistry Advanced Inorganic Chemistry - Volume IIS.
Chand Publishing

This book is specially designed for B.Sc. Chemistry Honours Degree students. However, it is believed to be helpful to post-graduate students also. It covers by and large physical chemistry part of the Chemistry Honours syllabus taught in different Indian Universities. Elaborate and lucid discussion of each chapter is the strength of this book. Questions and numerical problems are also included at the end of almost every chapter. Strenuous effort has been given to derive different mathematical equations as well as to handle quantum mechanics using mathematics taught in undergraduate level. The book contains 20 chapters, covering the following topics: - Thermodynamics is thoroughly discussed in this book, covering 1st law, 2nd law and 3rd law of thermodynamics, their applications, thermochemistry and its applications. Applications of thermodynamics in different areas like refrigerators, compressors, power plants, IC engines etc. are also discussed. Statistical thermodynamics is also discussed elaborately. - Chemical kinetics is another important part of chemistry since it covers reaction rate, order of a reaction, theory behind the reaction rate etc. Catalyst is also an important aspect since it has profound influence on reaction rate. Type of catalyst and mechanism of different catalyzed reactions are discussed in detail. A chemical reaction reaches an equilibrium state if carried out in a closed container. However, the equilibrium is sufficiently influenced by other parameters, like pressure, temperature etc. - Different physical states of matter (gaseous state, liquid state and solid state). In the solid state behavior of conductors and semiconductors are discussed thoroughly using quantum mechanics. - Detailed discussion of electrochemistry, electrochemical cell and

ionic equilibria is another important aspect of this book. Application of thermodynamics in electrochemical cell is also discussed. Concept of buffer solutions, pH and indicators are discussed in detail. - Phase equilibria is another important part of physical chemistry. The chapter includes details of phase rule, phase diagram, applications, different types of heterogeneous equilibrium system etc. - Colligative properties of dilute solutions are well documented, covering, Henry's law, Raoult's law of lowering of vapour pressure, elevation of boiling point, depression of freezing point, osmotic pressure etc. - Surface chemistry and properties of colloidal solutions are very much important in different chemical industries. These two sections are well discussed in this book. It includes details of derivation of different laws, theories behind the adsorption, stability of colloidal solutions etc. - Nuclear reactions are different from chemical reactions and energy, related to nuclear reactions is enormous, much higher than any chemical reaction. Study of different nuclear reactions including natural radioactivity, artificial radioactivity etc. and kinetics of nuclear reactions are well discussed in this book. Different areas of applications of nuclear reactions are also covered in this book. - Another important aspect of chemical reactions is chemical bonding. The book covers details of covalent bonding including quantum numbers, overlapping of atomic orbitals, molecular orbitals. Besides that ionic bonding and other types of bonding are also discussed in detail. - Photochemical reactions are different from chemical reactions. Light energy is the main source of photochemical reactions. Details of it including photochemical laws, mechanism etc. are well documented in this book.

GEORGE CHRISTOU Indiana University, Bloomington I am no doubt representative of a large number of current inorganic chemists in having obtained my undergraduate and postgraduate degrees in the 1970s. It was during this period that I began my continuing love affair with this subject, and the fact that it happened while I was a student in an organic laboratory is beside the point. I was always enchanted by the more physical aspects of inorganic chemistry; while being captivated from an early stage by the synthetic side, and the measure of creation with a small c that it entails, I nevertheless found the application of various theoretical, spectroscopic and physicochemical techniques to inorganic compounds to be fascinating, stimulating, educational and downright exciting. The various bonding theories, for example, and their use to explain or interpret spectroscopic observations were more or less universally accepted as belonging within the realm of inorganic chemistry, and textbooks of the day had whole sections on bonding theories, magnetism, kinetics, electron-transfer mechanisms and so on. However, things changed, and subsequent inorganic chemistry teaching texts tended to emphasize the more synthetic and descriptive side of the field. There are a number of reasons for this, and they no doubt include the rise of diamagnetic organometallic chemistry as the dominant subdiscipline within inorganic chemistry and its relative narrowness vis-d-vis physical methods required for its prosecution. An advanced-level textbook of inorganic chemistry for the graduate (B.Sc) and postgraduate (M.Sc) students of Indian and foreign universities. This book is a part of four volume series, entitled "A Textbook of Inorganic Chemistry – Volume I, II, III, IV".

CONTENTS: Chapter 1. Stereochemistry and Bonding in Main Group Compounds: VSEPR theory, $d^2 - p^2$ bonds, Bent rule and energetic of hybridization. Chapter 2. Metal-Ligand Equilibria in Solution: Stepwise and overall formation constants and their interactions, Trends in stepwise constants, Factors affecting stability of metal complexes with reference to the nature of metal ion and ligand, Chelate effect and its thermodynamic origin, Determination of binary formation constants by pH-metry and spectrophotometry. Chapter 3. Reaction Mechanism of Transition Metal Complexes – I: Inert and labile complexes, Mechanisms for ligand replacement reactions, Formation of complexes from aquo ions, Ligand displacement reactions in octahedral complexes- acid hydrolysis, Base hydrolysis, Racemization of tris chelate complexes, Electrophilic attack on ligands. Chapter 4. Reaction Mechanism of Transition Metal Complexes – II: Mechanism of ligand displacement reactions in square planar complexes, The trans effect, Theories of trans effect, Mechanism of electron transfer reactions – types; Outer sphere electron transfer mechanism and inner sphere electron transfer mechanism, Electron exchange. Chapter 5. Isopoly and Heteropoly Acids and Salts: Isopoly and Heteropoly acids and salts of Mo and W: structures of isopoly and heteropoly anions. Chapter 6. Crystal Structures: Structures of some binary and ternary compounds such as fluorite, antiferite, rutile, antirutile, cristobalite, layer lattices- CdI_2 , BiI_3 ; ReO_3 , Mn_2O_3 , corundum, perovskite, Ilmenite and Calcite. Chapter 7. Metal-Ligand Bonding: Limitation of crystal field theory, Molecular orbital theory, octahedral, tetrahedral or square planar complexes, π -bonding and molecular orbital theory. Chapter 8. Electronic Spectra of Transition Metal Complexes: Spectroscopic ground states, Correlation and spin-orbit coupling in free ions for 1st series of transition metals, Orgel and Tanabe-Sugano diagrams for transition metal complexes ($d^1 - d^9$ states), Calculation of Dq , B and β parameters, Effect of distortion on the d-orbital energy levels, Structural evidence from electronic spectrum, John-Teller effect, Spectrochemical and nephelauxetic series, Charge transfer spectra, Electronic spectra of molecular addition compounds. Chapter 9. Magnetic Properties of Transition Metal Complexes: Elementary theory of magneto-chemistry, Guoy's method for determination of magnetic susceptibility, Calculation of magnetic moments, Magnetic properties of free ions, Orbital contribution, effect of ligand-field, Application of magneto-chemistry in structure determination, Magnetic exchange coupling and spin state cross over. Chapter 10. Metal Clusters: Structure and bonding in higher boranes, Wade's rules, Carboranes, Metal Carbonyl Clusters - Low Nuclearity Carbonyl Clusters, Total Electron Count (TEC). Chapter 11. Metal- π Complexes: Metal carbonyls, structure and bonding, Vibrational spectra of metal carbonyls for bonding and structure elucidation, Important reactions of metal carbonyls; Preparation, bonding, structure and important reactions of transition metal nitrosyl, dinitrogen and dioxygen complexes; Tertiary phosphine as ligand.

Pratiyogita Darpan (monthly magazine) is India's largest read General Knowledge and Current Affairs Magazine. Pratiyogita Darpan (English monthly magazine) is known for quality content on General Knowledge and Current Affairs. Topics ranging from national and international news/ issues, personality development, interviews of examination toppers, articles/ write-up on topics like career, economy, history, public administration, geography, polity, social, environment, scientific, legal etc, solved papers of various examinations, Essay and debate contest, Quiz and knowledge testing features are covered every month in this magazine.

A Textbook for B.Sc. (Part III and Hons.) and Postgraduate Courses of Indian Universities. In this edition, I have made major changes in the light of modern concepts introduced in syllabi at the under-graduate and postgraduate level as well. With matter has also been updated. The subject matter has been arranged systematically, in a lucid style and simple language. New Problems and exercises have also been introduced to acquaint the students with trend of questions they expect in the examinations. Organic Spectroscopy presents the derivation of structural information from UV, IR, Raman, ^1H NMR, ^{13}C NMR, Mass and ESR spectral data in such a way that stimulates interest of students and researchers alike. The application of spectroscopy for structure determination and analysis has seen phenomenal growth and is now an integral part of Organic Chemistry courses. This book provides: -A logical, comprehensive, lucid and accurate presentation, thus making it easy to understand even through self-study; -Theoretical aspects of spectral techniques necessary for the interpretation of spectra; -Salient features of instrumentation involved in spectroscopic methods; -Useful spectral data in the form of tables, charts and figures; -Examples of spectra to familiarize the reader; -Many varied problems to help build competence and confidence; -A separate chapter on 'spectroscopic solutions of structural problems' to emphasize the utility of spectroscopy. Organic Spectroscopy is an invaluable reference for the interpretation of various spectra. It can be used as a basic text for undergraduate and postgraduate students of spectroscopy as well as a practical resource by research chemists. The book will be of interest to chemists and analysts in academia and industry, especially those engaged in the synthesis and analysis of organic compounds including drugs, drug intermediates, agrochemicals, polymers and dyes.

"Release any concerns that you have about this path you're about to walk on, and just go into it with the curiosity of a child, knowing that you'll come out on the other side as a new person, or to be more accurate, far more of the person who you truly are and have been created to be." Jack Valentine seemed to have it all. He made good money as an adman, and looked good doing it. He had a hot apartment, cool friends, even a slick car—at least until the hectic Monday morning a truck smashed into it, sending the critically injured Jack to the hospital. Everything happens for a reason, though, and Jack's reason reveals itself in the silver-haired cancer patient who becomes his roommate one evening. The elderly man, Cal, shares his life story—one not dissimilar to Jack's—of material wealth masking a gaping hole within. Cal ultimately found salvation through philosophy ("the love of wisdom"), and now offers to help Jack by prepping the younger man for the Final Questions we all must face: Have I lived wisely? Have I loved well? Have I served greatly? Presenting Jack with three plane tickets, each accompanied by a map marked with a red X, Cal sends Jack to meet with three great teachers, each of whom will help Jack answer one of the Final Questions—just as they once helped Cal. First, in Rome, Jack will meet "the Saint." Then a haunted beach in Hawaii introduces him to "the Surfer." And finally the grandeur of New York City sets the stage for his last encounter: with "the CEO." Along the way, Jack will learn to do his interior work, discover that our

negative traits offer gateways to higher versions of ourselves, and understand that figuring things out in your head can distract you from the powerful whispers of your heart. Join Jack on his journey and step into the you that you were always meant to be.

Written primarily to meet the requirements of students at the undergraduate level, this book aims for a self-learning approach. The fundamentals of physical chemistry have been explained with illustrations, diagrams, tables, experimental techniques and solved problems.

The Third Edition Of Quantum Chemistry Is A Fully Updated Textbook Covering The Model Syllabus For M.Sc General Course Recently Circulated By Ugc To All Indian Universities. The Book Contains The Developments That Led To The Evolution Of Quantum Mechanics As Well As The Basic Concepts Of Quantum Mechanical Formalism In As Simple Terms As Possible. The Exposition Of The Principles Is Followed By Application To Transnational Motion Of Micro Particles (With Infinite And Finite Barriers), Vibrational And Rotational Motions, Perturbation And Variation Methods Atomic Structure, Etc. The Ories Of Chemical Bond - Molecular Orbital And Valence Bond - In Diatomic As Well As Polyatomic Molecules Are Elaborately Expanded With Sufficient Examples. In Poly Electronic Atoms And Polyatomic Molecules, The Apparently Complicated Theories - Hfrscf, Configuration Interaction, Extended Huckel Theory, Etc. Are Presented With Utmost Clarity And Examples. The Chapter On Molecular Symmetry And Group Theory, Which Find Frequent Applications In Simplifying Problems Particularly In Mo Treatment, Is An Additional Feature. Steps Involved In Mathematical Derivations Are Presented In Full Leaving No Ambiguity. Illustrative Examples And Practice Problems, With Hints Provided, Are Given In Every Chapter. The Book May Prove To Be A Self-Educator.

Saraswati Health and Physical Education is a much acclaimed and popular series in Health and Physical Education. The series demonstrates a deep understanding of the principles and concepts related to the subject while providing students with all the pedagogical tools necessary for comprehension and application. The fully revised edition, which includes all the latest developments in the field, in its colourful avatar will not only enhance the teaching-learning process but will also make it more enjoyable.

Learning the basics of physical chemistry with a unique, innovative approach. Georg Job and Regina Rueffler introduce readers to an almost intuitive understanding of the two fundamental concepts, chemical potential and entropy. Avoiding complex mathematics, these concepts are illustrated with the help of numerous demonstration experiments. Using these concepts, the subjects of chemical equilibria, kinetics and electrochemistry are presented at an undergraduate level. The basic quantities and equations necessary for the qualitative and quantitative description of chemical transformations are introduced by using everyday experiences and particularly more than one hundred illustrative experiments, many presented online as videos. These are in

turn supplemented by nearly 400 figures, and by learning objectives for each chapter. From a review of the German edition: "This book is the most revolutionary textbook on physical chemistry that has been published in the last few decades."

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