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Thomas P. Edmonds

Telecourse Guide for Accounting in Action for Use with Financial Accounting Robert Libby, Patricia A. Libby, Daniel G. Short, 2000-12

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Introductory Financial Accounting for Business ISE Thomas Edmonds, Christopher Edmonds, Mark Edmonds, Jennifer Edmonds, Philip Olds, 2024-08-13

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web verified by toppr correct option is d atoms having same atomic number but different atomic mass are known as isotopes the difference in the mass is due to the difference in their number of neutrons with the same number of electrons and protons

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web part i answer the questions based on the above reading what is an isotope isotopes are versions of the same element they have the same number of

protons and electrons as the element but different mass numbers and number of neutrons what does the number next to isotopes signify the number indicates the isotope s mass number

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number of neutrons in the atom the actual structure doesn t differ all that much but its mass does differ thus its density is a little

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