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 moles of propane are combusted it is a 4 1 ration so the  
 moles of water 1 45 4 5 8 moles b how many molecules of  
 water is this remember each mole has  $6.02 \times 10^{23}$  molecules  
 $5.86 \times 10^{23}$   $3.49 \times 10^{24}$  2 if 2 35 moles of  $\text{CO}_2$  are produced  
 in a reaction how many moles of  
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