

# Java Concurrency In Practice

**Camilla Rothe**

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the following sets in order of increasing  
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following molecules are polar or  
nonpolar hcn h c n cf4 c f 4 cocl2 c o c l  
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shared and the bond is  
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contains polar bonds can be nonpolar  
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the following molecules and ions  
contain polar bonds which of these  
molecules and ions have dipole  
moments a clf 5 b clo 2 clo 2 c tecl2 4  
tecl 4 2 d pcl 3 e sef 4 f ph 2 ph 2 g xef  
2 answer problem 6 2 3 6 2 3

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the bonded atoms is h c cl and so the  
bond moments all point toward the cl  
end of the molecule and sum to yield a  
considerable dipole moment the  
molecules are relatively polar for  
molecules of high symmetry such as bf  
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