Handbook Of Aqueous Electrolyte Solutions Physical Properties Estimation And Correlation Methods Ellis Horwood Series In Physical Chemistry

The CRC Handbook of Thermodynamic Data of Aqueous Polymer Solutions provides a new and complete collection of the practical thermodynamic data required by researchers and engineers for a variety of applications including: basic and applied chemistry; chemical engineering; thermodynamic research; computational modeling; membrane science and technolo

Hydrometallurgy '94 contains the 78 papers that were presented at the international symposium organized by the Institution of Mining and Metallurgy and the Society of Chemical Industry and held in Cambridge, England, in July 1994. In the papers specific attention is paid to the concept of sustainable development and the associated ideas of cleaner technology, recycling and waste minimization that have particular relevance to the extractiona nd processing of metals and other mineral products. The papers, by authors from 30 contries, are grouped under the headings: Hydrometallurgy and Sustainable Development; Materials Production and the Environment; Fundamentals; Leaching; Bioprocessing; Gold Solution Purification; Effluent Treatment; Processes; and Recycling.

This book covers the fundamentals of the rapidly growing field of biothermodynamics, showing how thermodynamics can best be applied to applications and processes in biochemical engineering. It describes the rigorous application of thermodynamics in biochemical engineering to rationalize bioprocess development and obviate a substantial fraction of this need for tedious experimental work. As such, this book will appeal to a diverse group of readers, ranging from students and professors in biochemical engineering, to scientists and engineers, for whom it will be a valuable reference.

Demand for better reliability from drug delivery systems has caused designers and researchers to move away from trial-and-error approaches and toward model-based methods of product development. Developing such models requires cross-disciplinary physical, mathematical, and physiological knowledge. Combining these areas under a single cover, Understanding Drug Release and Absorption Mechanisms builds a firm understanding ofall elements needed to conceive, build, and implement successful models of drug release. Written by experts with broad industrial and academic experience, this book discusses the underlying physical principles, shows how to build mathematical models based on these principles, and finally compares the resulting models with experimental results. The authors begin by introducing the basics of modeling, physiological details of gastrointestinal and dermal absorption pathways, rheology, mass transport and thermodynamics, dissolution and partitioning, as well as size effects on the dissolution of crystallites. From this

baseline, the authors explore applications in drug release from various delivery systems, specifically matrix systems, microemulsions, and permeability through membranes. Working systematically from theory to working models, Understanding Drug Release and Absorption Mechanisms: A Physical and Mathematical Approach demonstrates the steps involved in designing, building, and implementing realistic and reliable models of drug release without unrealistically simplifying the theoretical parameters.

This book is concerned with the theoretical principles of hydrometallurgical processes and engineering aspects. The hydrometallurgical processes of production of copper are discussed and leaching of chalcopyrite as the main sulphide mineral of copper processed in industry is used as an example. The book is suitable as a university textbook for students of metallurgy. Examines the different techniques involved Discusses the production of specific metals using hydrometalluric processes Looks at the future of hydrometallurgy

This book was first published in 1991. It considers the concepts and theories relating to mostly aqueous systems of activity coefficients.

Volume six of the proceedings considers the application of chemical oxidation to environmental problems, particularly treating wastewater, groundwater, hazardous waste, and air. Among the 22 topics are the design of an advanced oxidation process for decolorizing reactive dye waste using Fenton's reagent, a comparison of chlorine and bromine for chemical oxidation and disinfection, chlorine dioxide for disinfecting secondary effluent,

oxidizing secondary alcohols and sulfides by halamine polymers, the wet-air oxidation of phenolic compounds, and the catalyzed chemical oxidation of VOCs. Reproduced from typescripts, many double spaced. No index. Annotation copyrighted by Book News, Inc., Portland, OR

Focus on Chemistry & Biochemistry

Provides a bibliography of more than three thousand handbooks in various aspects of science and technology, from abrasives and band structures to yield strength and zero defects

Surface tension provides a thermodynamic avenue for analyzing systems in equilibrium and formulating phenomenological explanations for the behavior of constituent molecules in the surface region. While there are extensive experimental observations and established ideas regarding desorption of ions from the surfaces of aqueous salt solutions, a more successful discussion of the theory has recently emerged, which allows the quantitative calculation of the distribution of ions in the surface region. Surface Tension and Related Thermodynamic Quantities of Aqueous Electrolyte Solutions provides a detailed and systematic analysis of the properties of ions at the air/water interface. Unifying older and newer theories and measurements, this book emphasizes the contributions of simple ions to surface tension behavior, and the practical consequences. It begins with a general discussion on Gibbs surface thermodynamics, offering a guide to his theoretical insight and formulation of the boundary between fluids. The text then discusses the thermodynamic formulae that are useful for practical experimental work in the analysis of fluid/fluid interfaces. Chapters cover surface tension of pure water at air/water and air/oil interfaces, surface tension of solutions and the

thermodynamic quantities associated with the adsorption and desorption of solutes, and surface tension of simple salt solutions. They also address adsorption of ions at the air/water interface, surface tension of solutions and the effect of temperature, adsorption from mixed electrolyte solutions, and thermodynamic properties of zwitterionic amino acids in the surface region. Focusing on the thermodynamic properties of ions at air/fluid interfaces, this book gives scientists a quantitative, rigorous, and objectively experimental methodology they can employ in their research.
Taking greater advantage of powerful computing capabilities over the last several years, the development of fundamental information and new models has led to major advances in nearly every aspect of chemical engineering. Albright's Chemical Engineering Handbook represents a reliable source of updated methods, applications, and fundamental concepts that will continue to play a significant role in driving new research and improving plant design and operations. Well-rounded, concise, and practical by design, this handbook collects valuable insight from an exceptional diversity of leaders in their respective specialties. Each chapter provides a clear review of basic information, case examples, and references to additional, more in-depth information. They explain essential principles, calculations, and issues relating to topics including reaction engineering, process control and design, waste disposal, and electrochemical and biochemical engineering. The final chapters cover aspects of patents and intellectual property, practical communication, and ethical considerations that are most relevant to engineers. From fundamentals to plant operations, Albright's Chemical Engineering Handbook offers a thorough, yet succinct guide to day-to-day methods and calculations used in chemical engineering applications. This handbook will serve the needs of practicing professionals as well as students preparing to

enter the field.
A unique text presenting practical information on the topic of nucleation and crystal growth processes from metastable solutions and melts Nucleation and Crystal Growth is a groundbreaking text thatoffers an overview and description of the processes and phenomena associated with metastability of solutions and melts. The author—a noted expert in the field—puts the emphasis on low-temperature solutions that are typically involved in crystallization in a wide range of industries. The text begins with a review of the basic knowledge of solutions and the fundamentals of crystallization processes. The author then explores topics related to the metastable state of solutions and melts from the standpoint of three-dimensional nucleation and crystal growth. Nucleation and Crystal Growth is the first text that contains a unified description and discussion of the many processes and phenomena occurring in the metastable zone of solutions and melts from the consideration of basic concepts of structure of crystallization. This important text: Outlines an interdisciplinary approach to the topic and offers an essential guide for crystal growth practitioners in materials science, physics, and chemical engineering Contains a comprehensive content that details the crystallization processes starting from the initial solutions and melts, all the way through nucleation, to the final crystal products Presents a unique focus and is the first book on understanding, and exploiting, metastability of solutions and melts in crystallization processes Written for specialists and researchers in the fields of materials science, condensed matter physics, and chemical engineering. Nucleation and Crystal Growth is a practical resource filled with hands-on knowledge of nucleation and crystal growth processes from metastable solutions and melts.
Properties of Aqueous Solutions of Electrolytes is a handbook

that systematizes the information on physico-chemical parameters of multicomponent aqueous electrolyte solutions. This important data collection will be invaluable for developing new methods for more efficient chemical technologies, choosing optimal solutions for more effective methods of using raw materials and energy resources, and other such activities. This edition, the first available in English, has been substantially revised and augmented. Many new tables have been added because of a significantly larger list of electrolytes and their properties (electrical conductivity, boiling and freezing points, pressure of saturated vapors, activity and diffusion coefficients). The book is divided into two sections. The first section provides tables that list the properties of binary aqueous solutions of electrolytes, while the second section deals with the methods for calculating their properties in multicomponent systems. All values are given in PSI units or fractional and multiple units. Metrological characteristics of the experimental methods used for the determination of physico-chemical parameters are indicated as a relative error and those of the computational methods as a relative error or a root-mean square deviation.
Nonaqueous Electrolytes Handbook, Volume II, is an authoritative and updated information source for nonaqueous solvent systems. The information in this handbook covers literature to 1973 and includes data for some 310 solvent systems. This volume has been organized to include 11 well-defined areas: Solubilities of Electrolytes; EMF and Potentiometric Titrations; Vapor Pressures; Cryoscopy; Heats of Solution Calorimetry; Polarography; Ligand Exchange Rates and Electrode Reactions; Electrical Double Layer; Spectroscopy and Structure of Electrolytes; Organic Electrolyte Battery Systems; and Additional References and Data Sources. The section on polarography is divided further according to inorganic electrolytes, organic electrolytes, and

organometallic compounds, in order to present the wealth of data in a concise and orderly manner. As in Volume I, the last section covers additional data sources, reviews, and data and references that were received too late to include in the earlier sections. The method of presentation of material is briefly described in the introduction to each section to facilitate the use of the tabulated information and bibliographies are given at the end of each section. A Compound Index is included.
Molecular Theory of Solvation presents the recent progress in the statistical mechanics of molecular liquids applied to the most intriguing problems in chemistry today, including chemical reactions, conformational stability of biomolecules, ion hydration, and electrode-solution interface. The continuum model of "solvation" has played a dominant role in describing chemical processes in solution during the last century. This book discards and replaces it completely with molecular theory taking proper account of chemical specificity of solvent. The main machinery employed here is the reference-interaction-site-model (RISM) theory, which is combined with other tools in theoretical chemistry and physics: the ab initio and density functional theories in quantum chemistry, the generalized Langevin theory, and the molecular simulation techniques. This book will be of benefit to graduate students and industrial scientists who are struggling to find a better way of accounting and/or predicting "solvation" properties.
This is the fifth volume in a series of books focusing on natural gas engineering, focusing on the extraction and disposal of acid gas. This volume includes information for both upstream and downstream operations, including chapters on modeling, carbon capture, chemical and thermodynamic models, and much more. Written by some of the most well-known and respected chemical and process engineers working with natural gas today, the chapters in this

important volume represent the most cutting-edge and state-of-the-art processes and operations being used in the field. Not available anywhere else, this volume is a must-have for any chemical engineer, chemist, or process engineer working with natural gas. There are updates of new technologies in other related areas of natural gas, in addition to the extraction and disposal of acid gas, including testing, reservoir simulations, acid gas injection, and natural gas hydrate formations. Advances in Natural Gas Engineering is an ongoing series of books meant to form the basis for the working library of any engineer working in natural gas today. Every volume is a must-have for any engineer or library.

This book is based on the undergraduate and MSc courses in hydrometallurgy which Professor A R Burkin gave from 1961 until he retired in 1988. It is divided into two sections. The first deals with the fundamental chemical and physical principles on which the technology is based. In the second, processes which are used for the production of individual metals are described, in terms of those principles where appropriate.

Chemical & Biochemical Kinetics - Mechanism of Reactions

Handbook of Aqueous Electrolyte SolutionsPhysical Properties, Estimation and Correlation MethodsHandbook of Aqueous Electrolyte ThermodynamicsTheory & ApplicationJohn Wiley & Sons

Nonaqueous Electrolytes Handbook, Volume I, is an authoritative and updated information source for nonaqueous solvent systems. The information in this handbook covers the literature to 1972 and includes data for some 210 solvents. The book has been organized into eight well-defined areas: Physical Properties o f

Solvents, Solvent Purification, Electrical Conductance, Diffusion, Density, Viscosity, Transference Numbers, and Additional References and Data Sources. The latter section covers additional data sources and reviews not adequately described in the preceding sections; recent data and references are also found in this section. The method of presentation of material is briefly described in the introduction to each section to facilitate the use of the tabulated information. Bibliographies are given at the end of each section. A Compound Index is included. Electrical conductance is the property most widely investigated. In view of the wealth of data, this section has been organized by solutes as follows: acids and alkali metal compounds, including ammonium compounds; quaternary ammonium salts and amines; solvent systems, electrolyte systems, and finally, all other solutes. For each, the data are reported not only for single component nonaqueous solvents but also for mixed solvents.

This book provides a thorough discussion of the thermodynamics of aqueous solutions and presents tools for analyzing and solving scientific and practical problems arising in this area. It also presents methods that can be used to deal with ionic and nonionic aqueous solutions under sub- or supercritical conditions. Illustrations and tables give examples of procedures employed to predict thermodynamic quantities of the solutions, and an appendix summarizing statistical mechanical equations used to describe the systems is also provided. High-Temperature Aqueous Solutions: Thermodynamic Properties contains essential

information for physical chemists, geochemists, geophysicists, chemical technicians, and scientists involved in electric power generation.

This new handbook will be an essential resource for ceramicists. It includes contributions from leading researchers around the world and includes sections on Basic Science of Advanced Ceramics, Functional Ceramics (electro-ceramics and optoelectro-ceramics) and engineering ceramics. Contributions from more than 50 leading researchers from around the world Covers basic science of advanced ceramics, functional ceramics (electro-ceramics and optoelectro-ceramics), and engineering ceramics Approximately 750 illustrations Electrochemistry plays a key role in a broad range of research and applied areas including the exploration of new inorganic and organic compounds, biochemical and biological systems, corrosion, energy applications involving fuel cells and solar cells, and nanoscale investigations. The Handbook of Electrochemistry serves as a source of electrochemical information, providing details of experimental considerations, representative calculations, and illustrations of the possibilities available in electrochemical experimentation. The book is divided into five parts: Fundamentals, Laboratory Practical, Techniques, Applications, and Data. The first section covers the fundamentals of electrochemistry which are essential for everyone working in the field, presenting an overview of electrochemical conventions, terminology, fundamental equations, and electrochemical cells, experiments, literature, textbooks, and specialized books. Part 2 focuses on the different laboratory aspects

of electrochemistry which is followed by a review of the various electrochemical techniques ranging from classical experiments to scanning electrochemical microscopy, electrogenerated chemiluminesence and spectroelectrochemistry. Applications of electrochemistry include electrode kinetic determinations, unique aspects of metal deposition, and electrochemistry in small places and at novel interfaces and these are detailed in Part 4. The remaining three chapters provide useful electrochemical data and information involving electrode potentials, diffusion coefficients, and methods used in measuring liquid junction potentials. * serves as a source of electrochemical information * includes useful electrochemical data and information involving electrode potentials, diffusion coefficients, and methods used in measuring liquid junction potentials * reviews electrochemical techniques (incl. scanning electrochemical microscopy, electrogenerated chemiluminesence and spectroelectrochemistry)

This compilation - the first of its kind - fills a real gap in the field of electrolyte data. Virtually all self-diffusion data in electrolyte solutions as reported in the literature have been examined and the book contains over 400 tables covering diffusion in binary and ternary aqueous solutions, in mixed solvents, and of non-electrolytes in various solvents. An important feature of the compilation is that all data have been critically examined and their accuracy assessed. Other features are an introductory chapter in which the methods of measurement are reviewed; appendices containing tables of the limiting self-diffusion coefficients of ions; and a list of references

to data which have been omitted but where information about the diffusing system is given. This is the only complete compilation of self-diffusion data in electrolyte solutions. It will appeal to electrochemists in general, particularly now that recent developments in the theory of transport processes require these data. It will also have a special appeal to electroanalytical chemists in that the ionic self-diffusion coefficient is an important quantity for the interpretation of electrode reactions. In addition, the book will interest geochemists and environmental chemists because the migration of radioactive ions from nuclear waste in certain aqueous media will be governed by the tracer-diffusion coefficient.

Electrochemically Enabled Sustainability: Devices, Materials and Mechanisms for Energy Conversion covers topics related to current research in electrochemical power sources, highlighting some of the latest concepts in electrochemical conversion for sustainability. The book examines the most recent and innovative technologies employed in battery and fuel cell technology. It introduces the fundamental concepts applied to these electrochemical power sources and provides in-depth discussion on the materials, design, and performance of these devices. Written by internationally acclaimed experts, the chapters illustrate how key technologies for sustainability are enabled by electrochemical conversion. Topics include the reduction of carbon dioxide to resolve issues of carbon capture, energy storage, and generation of portable fuel; turning waste into energy using microbial fuel cells; the promise of vanadium redox flow batteries for massive energy storage; and improved performance of hybrid devices. The book addresses numerous aspects of lithium-type batteries for vehicle

propulsion and energy storage, presenting a broad range of lithium batteries, and considering nano-structuring issues, layered-structure materials, and hierarchical structure. This book provides timely coverage of critical issues in emerging and conventional technologies, presenting a wide range of electrochemical devices, related materials, and operation mechanisms. It stimulates an appreciation for the novelty of these electrochemical power sources and offers a projection of future integration of these devices in practice.

An Introduction to Aqueous Electrolyte Solutions is a comprehensive coverage of solution equilibria and properties of aqueous ionic solutions. Acid/base equilibria, ion pairing, complex formation, solubilities, reversible emf?s and experimental conductance studies are all illustrated by many worked examples. Theories of non-ideality leading to expressions for activity coefficients, conductance theories and investigations of solvation are described; great care being taken to provide detailed verbal clarification of the key concepts of these theories. The theoretical development focuses on the physical aspects, with the mathematical development being fully explained. An overview of the thermodynamic background is given. Each chapter includes intended learning outcomes and worked problems and examples to encourage student understanding of this multidisciplinary subject. An invaluable text for students taking courses in chemistry and chemical engineering. This book will also be useful for biology, biochemistry and biophysics students who may be required to study electrochemistry as part of their course. A comprehensive introduction to the behaviour and properties of aqueous ionic solutions, including clear explanation and development of key concepts and theories Clear, student friendly style clarifying complex aspects which students find difficult Key developments in concepts and theory explained in a descriptive manner to

encourage student understanding Includes worked problems and examples throughout

In this new textbook on physical chemistry, fundamentals are introduced simply yet in more depth than is common. Topics are arranged in a progressive pattern, with simpler theory early and more complicated theory later. General principles are induced from key experimental results. Some mathematical background is supplied where it would be helpful. Each chapter includes worked-out examples and numerous references. Extensive problems, review, and discussion questions are included for each chapter. More detail than is common is devoted to the nature of work and heat and how they differ. Introductory Caratheodory theory and the standard integrating factor for dGrev are carefully developed. The fundamental role played by uncertainty and symmetry in quantum mechanics is emphasized. In chemical kinetics, various methods for determined rate laws are presented. The key mechanisms are detailed. Considerable statistical mechanics and reaction rate theory are then surveyed. Professor Duffey has given us a most readable, easily followed text in physical chemistry.

This interdisciplinary and accessible new volume presents a broad range of application-based green chemistry and engineering research. The book familiarizes readers with the integration of tools and spell out the approaches for green engineering of new processes as well as improving the environmental risks of existing processes. The expert authors discuss the myriad opportunities and the challenges facing green chemistry today in both its theoretical and practical implementation. The book expands upon green chemistry concepts with the latest research and new and innovative applications, providing both the breadth and depth researchers need. Topics include solar energy, electrospinning of bio-based polymeric nanofibers,

biotransformation, engineered nanomaterials in environmental protection, and much more.

Crystallization is an important separation and purification process used in industries ranging from bulk commodity chemicals to specialty chemicals and pharmaceuticals. In recent years, a number of environmental applications have also come to rely on crystallization in waste treatment and recycling processes. The authors provide an introduction to the field of newcomers and a reference to those involved in the various aspects of industrial crystallization. It is a complete volume covering all aspects of industrial crystallization, including material related to both fundamentals and applications. This new edition presents detailed material on crystallization of biomolecules, precipitation, impurity-crystal interactions, solubility, and design. Provides an ideal introduction for industrial crystallization newcomers Serves as a worthwhile reference to anyone involved in the field Covers all aspects of industrial crystallization in a single, complete volume

Volume 17 in the Ion Exchange and Solvent Extraction series represents the vanguard of research on solvent extraction. It covers the principles of electrolyte extraction and other subjects of increasing interest to the field. This volume begins with pharmaceutical applications of supercritical fluid solvents, particularly supercritical carbon dioxide. It also contains chapters on liquid ion exchangers and relevant experiment protocols, SCF applications in drug formulation and pollution reduction, exploiting SCF as reaction media, applications of metal bis(dicarbollide) in analytical chemistry and radioactive waste treatment, and synergistic extraction of metal ions. Volume 17 discusses the ion exchange isothermal supersaturation technique, metal separation via pH-induced parametric pumping, modeling of ion exchange kinetics for ultrapure water, and the engineering of activated carbons and

carbonaceous materials for removal of metal ions and organic micropollutants in water. Volume 17 cover topics that include supercritical fluid applications, applications of metal bis(dicarbollide), and synergistic extraction of metal ions.
A UNIQUE BOOK ON THE PRESENT STATUS OF SOLVENTS AND SOLUTIONS WITH IMPORTANT PROBLEMS RELATED TO THEIR STRUCTURE AND PROPERTIES The literature on the properties of solvents and solutions used in academic research and in a wide range of industries has grown enormously during the last four decades, and is scattered in different specialized journals. Solvents and Solutions is a groundbreaking text that offers a systematic compilation of important problems related to selected properties of solvents and solutions based on the literature published so far. The author places emphasis on explaining the basic concepts involved in understanding the properties and behavior of various solvents and solutions of electrolytes and nonelectolytes in a consistent manner. After a description of the general characteristics of structure of solvents and solutions and the solubility of electrolytes and nonelectrolytes under normal temperature and pressure conditions, the book first deals with different aspects of the density and the refractive index of solvents and dilute as well as concentrated solutions, and finally with the transport (i.e. viscosity and electric conductivity) and thermal properties of solvents and solutions. Solvents and solutions is the first text devoted to the description and discussion of their properties since the publication of a monograph on the physical properties of aqueous electrolyte solutions more than three decades ago. The main features of this book are: Reflects developments in the investigation of solvents and solutions during the last three decades. Outlines basic concepts involved in understanding the properties and behavior of solvents and solutions. Describes and discusses different

properties of ionic liquids as solvents and the behavior of their mixtures with other commonly used solvents. Contents of different chapters are not only self-contained but the contents are practically independent of each other. Written as a practical guide for researchers who are looking for an uptodate overview of the physical and transport properties of solvents and solutions, and as a reference source for workers in chemical industries and related fields and for graduate students of chemical engineering and physical chemistry. Food properties, whether they concern the physical, thermodynamic, chemical, nutritional or sensory characteristics of foods, play an important role in food processing. In our quest to gain a mechanistic understanding of changes occurring during food processing, the knowledge of food properties is essential. Quantitative information on the food properties is necessary in the design and operation of food processing equipment. Foods, because of their biological nature and variability, vary in the magnitude of their properties. The variation in properties offer a challenge both in their measurement and use in the food processing applications. Often a high level of precision in measurement of properties is not possible as the measurement method may itself cause changes to the product, resulting in a variation in the obtained values. Recognizing the difficulties in measurement of food properties, and the lack of completeness of such information, several research programs have been in existence during the last two decades. In Europe, a multinational effort has been underway since 1978. The first project supported by COST (European Cooperation in the Field of Scientific and Technical Research), was titled COST 90 "The Effect of Processing on the Physical Properties of Foodstuffs". This and another project COST 90bis have considerably added to our knowledge of measurement methods and data on a number of physical

properties. Two publications that summarize the work conducted under 1 2 these projects are Physical Properties of Foods and Physical Properties of Foods .

Expertise in electrolyte systems has become increasingly important in traditional CPI operations, as well as in oil/gas exploration and production. This book is the source for predicting electrolyte systems behavior, an indispensable "do-it-yourself" guide, with a blueprint for formulating predictive mathematical electrolyte models, recommended tabular values to use in these models, and annotated bibliographies. The final chapter is a general recipe for formulating complete predictive models for electrolytes, along with a series of worked illustrative examples. It can serve as a useful research and application tool for the practicing process engineer, and as a textbook for the chemical engineering student.

In this series Rajiv Kohli and Kash Mittal have brought together the work of experts from different industry sectors and backgrounds to provide a state-of-the-art survey and best-practice guidance for scientists and engineers engaged in surface cleaning or handling the consequences of surface contamination. The expert contributions in this volume cover important fundamental aspects of surface contamination that are key to understanding the behavior of specific types of contaminants. This understanding is essential to develop preventative

and mitigation methods for contamination control. The coverage complements the treatment of surface contamination in vol.1, Fundamental and Applied Aspects. This volume covers: Sources and Generation of Particles; Manipulation Techniques for Particles on Surfaces; Particle Deposition and Rebound; Particle Behavior in Liquid Systems; Biological and Metallic Contamination; and includes a comprehensive list of current standards and resources. Feature: Comprehensive coverage of innovations in surface contamination and cleaning Benefit: One-stop series where a wide range of readers will be sure to find a solution to their cleaning problem, saving the time involved in consulting a range of disparate sources. Feature: Written by established experts in the contamination and cleaning field Benefit: Provides an authoritative resource Feature: Each chapter is a comprehensive review of the state of the art. Benefit: Can be relied on to provide insight, clarity and real expertise on up-to-the-minute innovations. Feature: Case studies included Benefit: Case studies help the reader see theory applied to the solution of real-world practical cleaning and contamination problems.

Reference Electrodes are a crucial part of any electrochemical system, yet an up-to-date and comprehensive handbook is long overdue. Here, an experienced team of electrochemists provides an in-depth source of information and data for the proper

choice and construction of reference electrodes. This includes all kinds of applications such as aqueous and non-aqueous solutions, ionic liquids, glass melts, solid electrolyte systems, and membrane electrodes. Advanced technologies such as miniaturized, conducting-polymer-based, screen-printed or disposable reference electrodes are also covered. Essential know-how is clearly presented and illustrated with almost 200 figures.

Optical and Molecular Physics: Theoretical Principles and Experimental Methods addresses many important applications and advances in the field. This book is divided into 5 sections: Plasmonics and carbon dots physics with applications Optical films, fibers, and materials Optical properties of advanced materials Molecular physics and diffusion Macromolecular physics Weaving together science and engineering, this new volume addresses important applications and advances in optical and molecular physics. It covers plasmonics and carbon dots physics with applications; optical films, fibers, and materials; optical properties of advanced materials; molecular physics and diffusion; and macromolecular physics. This book looks at optical materials in the development of composite materials for the functionalization of glass, ceramic, and polymeric substrates to interact with electromagnetic radiation and presents state-of-the-art research in preparation

methods, optical characterization, and usage of optical materials and devices in various photonic fields. The authors discuss devices and technologies used by the electronics, magnetics, and photonics industries and offer perspectives on the manufacturing technologies used in device fabrication.

As the various disciplines of science advance, they proliferate and tend to become more esoteric. Barriers of specialized terminologies form, which cause scientists to lose contact with their colleagues, and differences in points-of-view emerge which hinder the unification of knowledge among the various disciplines, and even within a given discipline. As a result, the scientist, and especially the student, is in many instances offered fragmented glimpses of subjects that are funda mentally synthetic and that should be treated in their own right. Such seems to be the case of the liquid state. Unlike the other states of matter -- gases, solids, and plasmas -- the liquid state has not yet received unified treatment, probably because it has been the least explored and remains the least understood state of matter. Occasionally, events occur which help remove some of the barriers that separate scientists and disciplines alike. Such an event was the ASI on The Liquid State held this past July at the lovely Hotel Tivoli Sintra, in the picturesque town of Sintra, Portugal, approximately 30 km northwest of

Lisbon. Since this broad a subject could not be covered in one Institute, the focus of the ASI was on a theme that provided a common thread of understanding for all in attendance -- the Electrical Proper ties of the Liquid State.

Thermodynamics of Phase Equilibria in Food Engineering is the definitive book on thermodynamics of equilibrium applied to food engineering. Food is a complex matrix consisting of different groups of compounds divided into macronutrients (lipids, carbohydrates, and proteins), and micronutrients (vitamins, minerals, and phytochemicals). The quality characteristics of food products associated with the sensorial, physical and microbiological attributes are directly related to the thermodynamic properties of specific compounds and complexes that are formed during processing or by the action of diverse interventions, such as the environment, biochemical reactions, and others. In addition, in obtaining bioactive substances using separation processes, the knowledge of phase equilibria of food systems is essential to provide an efficient separation, with a low cost in the process and high selectivity in the recovery of the desired component. This book combines theory and application of phase equilibria data of systems containing food compounds to help food engineers and researchers to solve complex problems found in food processing. It provides support to researchers

from academia and industry to better understand the behavior of food materials in the face of processing effects, and to develop ways to improve the quality of the food products. Presents the fundamentals of phase equilibria in the food industry Describes both classic and advanced models, including cubic equations of state and activity coefficient Encompasses distillation, solid-liquid extraction, liquid-liquid extraction, adsorption, crystallization and supercritical fluid extraction Explores equilibrium in advanced systems, including colloidal, electrolyte and protein systems