

Classical And Statistical Thermodynamics Solutions

Exceptionally articulate treatment of negative temperatures, relativistic effects, black hole thermodynamics, gravitational collapse, much more. Over 100 problems with worked solutions. Geared toward advanced undergraduates and graduate students.

Functionalized Polysulfones: Synthesis, Characterization, and Applications focuses on polysulfones and their derivatives, which are widely used as functional materials in the biochemical, industrial, and medical fields due to their structural and physical characteristics, such as good optical properties, high thermal and chemical stability, mechanical strength, resistance to extreme pH values, and low creep. Because of their antimicrobial actions, solubility characteristics, water permeability, and separation, the functional groups, which modify the hydrophilicity of polysulfones, are of particular interest for biomedical applications. In addition, the functional groups are an intrinsic requirement for affinity, ion exchange, and other special membranes. In this book, the bioapplications of polysulfones are presented in two categories: blood-contacting devices (e.g., membranes for hemodialysis, hemodiafiltration, and hemofiltration) and cell- or tissue-contacting devices (e.g., bioreactors made by hollow-fiber membrane and nerve generation through polysulfone semipermeable hollow membrane). Surface wettability and hydrophilicity trends, as well as the morphological characteristics of modified polysulfones, are analyzed for semipermeable membrane purposes. Select chapters provide an introduction to chelating units on the modified polysulfone structure to obtain potential applications, such as surface coatings on metals and glasses, adhesives, high-temperature lubricants, electrical insulators, semiconductors, and the reduction of heavy-metal

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pollution in ecosystems. Featuring recent scientific information, *Functionalized Polysulfones: Synthesis, Characterization, and Applications* advances the basic knowledge of students and researchers working in the field of polymeric materials, including physicists, chemists, engineers, bioengineers, and biologists.

This text presents statistical mechanics and thermodynamics as a theoretically integrated field of study. It stresses deep coverage of fundamentals, providing a natural foundation for advanced topics. The large problem sets (with solutions for teachers) include many computational problems to advance student understanding.

This Is An Introductory Book Which Explains The Foundations Of The Subject And Its Application. It Is Intended Primarily For Graduate Students But May Provide Useful Information And Reading To Science And Engineering Students At All Levels. It Assumes That Readers Have Knowledge Of Basic Thermodynamics And Quantum Mechanics. With This, The Theory Has Been Developed In A Simple, Logical And Understandable Way. Some Applications Of Statistical Thermodynamics Have Been Described In Detail With Illustrative Solved Examples. There Are Two Basic Approaches In Statistical Mechanics; One Based On The Study Of Independent Particles In An Isolated System And The Other Based On The Concept Of Ensembles. In This Book Attempt Has Been Made To Take Advantage Of Both Approaches. While The Fundamental Concepts Have Been Developed By First Approach, Concept Of Ensembles Have Been Included To Bring Out The Importance Of This Concept In The Application Of Statistical Thermodynamics To Chemical Systems Where Interparticle Interactions Become Important. Part I Of The Book Deals With The Background Concepts, Fundamentals In Mathematics, Classical Mechanics, Quantum Mechanics And

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Thermodynamics Which Are Essential For Statistical Mechanics. Part Ii Covers Formalism Of Statistical Mechanism And Its Relation To Thermodynamics As Well As The Statistical Mechanics Of Ensembles, Quantum Statistics And Fluctuations. Part Iii Includes Chapters On The Applications Of The Formalism To Real Laboratory Chemical Systems. In This Part Additions Such As Imperfect Gases, Equilibrium Isotope And Kinetic Isotope Effects And Reactions At The Surfaces Have Been Made, In This Edition. Part Iv Is Also An Addition Which Covers Quantum Systems Such As Ideal Fermi Gas (Free Electrons In Metals), Photon Gas And Ideal Bose Gas (Helium Gas).

This book provides a solid introduction to the classical and statistical theories of thermodynamics while assuming no background beyond general physics and advanced calculus. Though an acquaintance with probability and statistics is helpful, it is not necessary. Providing a thorough, yet concise treatment of the phenomenological basis of thermal physics followed by a presentation of the statistical theory, this book presupposes no exposure to statistics or quantum mechanics. It covers several important topics, including a mathematically sound presentation of classical thermodynamics; the kinetic theory of gases including transport processes; and thorough, modern treatment of the thermodynamics of magnetism. It includes up-to-date examples of applications of the statistical theory, such as Bose-Einstein condensation, population inversions, and white dwarf stars. And, it also includes a chapter on the connection between thermodynamics and information theory. Standard International units are used throughout. An important reference book for every professional whose work requires and understanding of thermodynamics: from engineers to industrial designers. This text explores the connections between different thermodynamic subjects related to fluid

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systems. Emphasis is placed on the clarification of concepts by returning to the conceptual foundation of thermodynamics and special effort is directed to the use of a simple nomenclature and algebra. The book presents the structural elements of classical thermodynamics of fluid systems, covers the treatment of mixtures, and shows via examples and references both the usefulness and the limitations of classical thermodynamics for the treatment of practical problems related to fluid systems. It also includes diverse selected topics of interest to researchers and advanced students and four practical appendices, including an introduction to material balances and step-by-step procedures for using the Virial EOS and the PRSV EOS for fugacities and the ASOG-KT group method for activity coefficients. The Olivera-Fuentes table of PRSV parameters for more than 800 chemical compounds and the Gmehling-Tochigi tables of ASOG interaction parameters for 43 groups are included.

Clearly connects macroscopic and microscopic thermodynamics and explains non-equilibrium behavior in kinetic theory and chemical kinetics.

Statistical thermodynamics and the related domains of statistical physics and quantum mechanics are very important in many fields of research, including plasmas, rarefied gas dynamics, nuclear systems, lasers, semiconductors, superconductivity, ortho- and para-hydrogen, liquid helium, and so on. *Statistical Thermodynamics: Understanding the Properties of Macroscopic Systems* provides a detailed overview of how to apply statistical principles to obtain the physical and thermodynamic properties of macroscopic systems. Intended for physics, chemistry, and other science students at the graduate level, the book starts with fundamental principles of statistical physics, before diving into thermodynamics. Going further than many advanced textbooks, it includes Bose-Einstein, Fermi-Dirac statistics, and Lattice

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dynamics as well as applications in polaron theory, electronic gas in a magnetic field, thermodynamics of dielectrics, and magnetic materials in a magnetic field. The book concludes with an examination of statistical thermodynamics using functional integration and Feynman path integrals, and includes a wide range of problems with solutions that explain the theory. Nobel Laureate's brilliant attempt to develop a simple, unified standard method of dealing with all cases of statistical thermodynamics — classical, quantum, Bose-Einstein, Fermi-Dirac, etc.

The statistical mechanical theory of liquids and solutions is a fundamental area of physical sciences with important implications for many industrial applications. This book shows how you can start from basic laws for the interactions and motions of microscopic particles and calculate how macroscopic systems of these particles behave, thereby explaining properties of matter at the scale that we perceive. Using this microscopic, molecular approach, the text emphasizes clarity of physical explanations for phenomena and mechanisms relevant to fluids, addressing the structure and behavior of liquids and solutions under various conditions. A notable feature is the author's treatment of forces between particles that include nanoparticles, macroparticles, and surfaces. The book also provides an expanded, in-depth treatment of polar liquids and electrolytes.

The aim of this book is to present Classical Thermodynamics in a unified way, from the most fundamental principles to non-uniform systems, thereby requiring the introduction

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of coarse graining methods, leading for instance to phase field methods. Solutions thermodynamics and temperature-concentration phase diagrams are covered, plus also a brief introduction to statistical thermodynamics and topological disorder. The Landau theory is included along with a general treatment of multicomponent instabilities in various types of thermodynamic applications, including phase separation and order-disorder transitions. Nucleation theory and spinodal decomposition are presented as extreme cases of a single approach involving the all-important role of fluctuations. In this way, it is hoped that this coverage will reconcile in a unified manner techniques generally presented separately in physics and materials texts.

In order to equip hopeful graduate students with the knowledge necessary to pass the qualifying examination, the authors have assembled and solved standard and original problems from major American universities – Boston University, University of Chicago, University of Colorado at Boulder, Columbia, University of Maryland, University of Michigan, Michigan State, Michigan Tech, MIT, Princeton, Rutgers, Stanford, Stony Brook, University of Tennessee at Knoxville, and the University of Wisconsin at Madison – and Moscow Institute of Physics and Technology. A wide range of material is covered and comparisons are made between similar problems of different schools to provide the student with enough information to feel comfortable and confident at the exam. Guide to Physics Problems is published in two volumes: this book, Part 2, covers Thermodynamics, Statistical Mechanics and Quantum Mechanics; Part 1, covers

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Mechanics, Relativity and Electrodynamics. Praise for A Guide to Physics Problems: Part 2: Thermodynamics, Statistical Physics, and Quantum Mechanics: "... A Guide to Physics Problems, Part 2 not only serves an important function, but is a pleasure to read. By selecting problems from different universities and even different scientific cultures, the authors have effectively avoided a one-sided approach to physics. All the problems are good, some are very interesting, some positively intriguing, a few are crazy; but all of them stimulate the reader to think about physics, not merely to train you to pass an exam. I personally received considerable pleasure in working the problems, and I would guess that anyone who wants to be a professional physicist would experience similar enjoyment. ... This book will be a great help to students and professors, as well as a source of pleasure and enjoyment." (From Foreword by Max Dresden) "An excellent resource for graduate students in physics and, one expects, also for their teachers." (Daniel Kleppner, Lester Wolfe Professor of Physics Emeritus, MIT) "A nice selection of problems ... Thought-provoking, entertaining, and just plain fun to solve." (Giovanni Vignale, Department of Physics and Astronomy, University of Missouri at Columbia) "Interesting indeed and enjoyable. The problems are ingenious and their solutions very informative. I would certainly recommend it to all graduate students and physicists in general ... Particularly useful for teachers who would like to think about problems to present in their course." (Joel Lebowitz, Rutgers University) "A very thoroughly assembled, interesting set of problems that covers the key areas of

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physics addressed by Ph.D. qualifying exams. ... Will prove most useful to both faculty and students. Indeed, I plan to use this material as a source of examples and illustrations that will be worked into my lectures." (Douglas Mills, University of California at Irvine)

Statistical Mechanics: Problems with solutions contains detailed model solutions to the exercise problems formulated in the companion Lecture notes volume. In many cases, the solutions include result discussions that enhance the lecture material. For readers' convenience, the problem assignments are reproduced in this volume.

A thorough understanding of statistical mechanics depends strongly on the insights and manipulative skills that are acquired through the solving of problems. Problems on Statistical Mechanics provides over 120 problems with model solutions, illustrating both basic principles and applications that range from solid-state physics to cosmology. An introductory chapter provides a summary of the basic concepts and results that are needed to tackle the problems, and also serves to establish the notation that is used throughout the book. The problems themselves occupy five chapters, progressing from the simpler aspects of thermodynamics and equilibrium statistical ensembles to the more challenging ideas associated with strongly interacting systems and nonequilibrium processes. Comprehensive solutions to all of the problems are designed to illustrate efficient and elegant problem-solving techniques. Where appropriate, the authors incorporate extended discussions of the points of principle that arise in the course of

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the solutions. The appendix provides useful mathematical formulae.

Volume 5.

"This textbook addresses the key questions in both classical thermodynamics and statistical thermodynamics: Why are the thermodynamic properties of a nano-sized system different from those of a macroscopic system of the same substance? Why and how is entropy defined in thermodynamics, and how is the entropy change calculated when dissipative heat is involved? What is an ensemble and why is its theory so successful?" "Translated from a highly successful Chinese book, this expanded English edition contains many updated sections and several new ones. They include the introduction of the grand canonical ensemble, the grand partition function and its application to ideal quantum gases, a discussion of the mean field theory of the Ising model and the phenomenon of ferromagnetism, as well as a more detailed discussion of ideal quantum gases near $T = 0$, for both Fermi and Bose gases."--BOOK JACKET.

Thermodynamic Properties of Nonelectrolyte Solutions reviews several of the more classical theories on the thermodynamics of nonelectrolyte solutions. Basic thermodynamic principles are discussed, along with predictive methods and molecular thermodynamics. This book is comprised of 12 chapters; the first of which introduces the reader to mathematical relationships, such as concentration variables, homogeneous functions, Euler's theorem, exact differentials, and method of least squares. The discussion then turns to partial molar quantities,

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ideal and nonideal solutions, and empirical expressions for predicting the thermodynamic properties of multicomponent mixtures from binary data. The chapters that follow explore binary and ternary mixtures containing only nonspecific interactions; the thermodynamic excess properties of liquid mixtures and ternary alcohol-hydrocarbon systems; and solubility behavior of nonelectrolytes. This book concludes with a chapter describing the use of gas-liquid chromatography in determining the activity coefficients of liquid mixtures and mixed virial coefficients of gaseous mixtures. This text is intended primarily for professional chemists and researchers, and is invaluable to students in chemistry or chemical engineering who have background in physical chemistry and classical thermodynamics.

Well respected and widely used, this volume presents problems and full solutions related to a wide range of topics in thermodynamics, statistical physics, and statistical mechanics. The text is intended for instructors, undergraduates, and graduate students of mathematics, physics, chemistry, and engineering. Twenty-eight chapters, each prepared by an expert, proceed from simpler to more difficult subjects. Similarly, the early chapters are easier than the later ones, making the book ideal for independent study. Subjects begin with the laws of thermodynamics and statistical theory of information and of ensembles,

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advancing to the ideal classical gases of polyatomic molecules, non-electrolyte liquids and solutions, and surfaces. Subsequent chapters explore imperfect classical and quantum gas, phase transitions, cooperative phenomena, Green function methods, the plasma, transport in gases and metals, Nyquist's theorem and its generalizations, stochastic methods, and many other topics.

Statistical Mechanics: Fundamentals and Model Solutions, Second Edition Fully updated throughout and with new chapters on the Mayer expansion for classical gases and on cluster expansion for lattice models, this new edition of Statistical Mechanics: Fundamentals and Model Solutions provides a comprehensive introduction to equilibrium statistical mechanics for advanced undergraduate and graduate students of mathematics and physics. The author presents a fresh approach to the subject, setting out the basic assumptions clearly and emphasizing the importance of the thermodynamic limit and the role of convexity. With problems and solutions, the book clearly explains the role of models for physical systems, and discusses and solves various models. An understanding of these models is of increasing importance as they have proved to have applications in many areas of mathematics and physics. Features Updated throughout with new content from the field An established and well-loved textbook Contains new problems and solutions for further learning opportunity

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Problems and Solutions on Thermodynamics and Statistical Mechanics
World Scientific

Statistical physics has its origins in attempts to describe the thermal properties of matter in terms of its constituent particles, and has played a fundamental role in the development of quantum mechanics. Based on lectures taught by Professor Kardar at MIT, this textbook introduces the central concepts and tools of statistical physics. It contains a chapter on probability and related issues such as the central limit theorem and information theory, and covers interacting particles, with an extensive description of the van der Waals equation and its derivation by mean field approximation. It also contains an integrated set of problems, with solutions to selected problems at the end of the book and a complete set of solutions is available to lecturers on a password protected website at www.cambridge.org/9780521873420. A companion volume, Statistical Physics of Fields, discusses non-mean field aspects of scaling and critical phenomena, through the perspective of renormalization group.

Publisher Description

An understanding of statistical thermodynamic molecular theory is fundamental to

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the appreciation of molecular solutions. This complex subject has been simplified by the authors with down-to-earth presentations of molecular theory. Using the potential distribution theorem (PDT) as the basis, the text provides a discussion of practical theories in conjunction with simulation results. The authors discuss the field in a concise and simple manner, illustrating the text with useful models of solution thermodynamics and numerous exercises. Modern quasi-chemical theories that permit statistical thermodynamic properties to be studied on the basis of electronic structure calculations are given extended development, as is the testing of those theoretical results with ab initio molecular dynamics simulations. The book is intended for students taking up research problems of molecular science in chemistry, chemical engineering, biochemistry, pharmaceutical chemistry, nanotechnology and biotechnology.

A comprehensive graduate textbook explaining key physical methods in biology, reflecting the very latest research in this fast-moving field.

This textbook covers the basic principles of statistical physics and thermodynamics. The text is pitched at the level equivalent to first-year graduate studies or advanced undergraduate studies. It presents the subject in a straightforward and lively manner. After reviewing the basic probability theory of classical thermodynamics, the author addresses the standard topics of statistical physics. The text demonstrates their relevance in other scientific fields using clear

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and explicit examples. Later chapters introduce phase transitions, critical phenomena and non-equilibrium phenomena.

Four-part treatment covers principles of quantum statistical mechanics, systems composed of independent molecules or other independent subsystems, and systems of interacting molecules, concluding with a consideration of quantum statistics.

Most interesting and difficult problems in equilibrium statistical mechanics concern models which exhibit phase transitions. For graduate students and more experienced researchers this book provides an invaluable reference source of approximate and exact solutions for a comprehensive range of such models. Part I contains background material on classical thermodynamics and statistical mechanics, together with a classification and survey of lattice models. The geometry of phase transitions is described and scaling theory is used to introduce critical exponents and scaling laws. An introduction is given to finite-size scaling, conformal invariance and Schramm—Loewner evolution. Part II contains accounts of classical mean-field methods. The parallels between Landau expansions and catastrophe theory are discussed and Ginzburg--Landau theory is introduced. The extension of mean-field theory to higher-orders is explored using the Kikuchi--Hijmans--De Boer hierarchy of approximations. In Part III the use of algebraic, transformation and decoration methods to obtain exact system information is considered. This is followed by an account of the use of transfer matrices for the location of incipient phase transitions in one-dimensionally infinite models and for exact solutions for two-dimensionally infinite systems. The latter is applied to a general analysis of eight-vertex models yielding as special cases the two-dimensional Ising model and the six-vertex model. The treatment of exact results ends with a discussion of dimer models. In Part IV series methods

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and real-space renormalization group transformations are discussed. The use of the De Neef—Enting finite-lattice method is described in detail and applied to the derivation of series for a number of model systems, in particular for the Potts model. The use of Padé, differential and algebraic approximants to locate and analyze second- and first-order transitions is described. The realization of the ideas of scaling theory by the renormalization group is presented together with treatments of various approximation schemes including phenomenological renormalization. Part V of the book contains a collection of mathematical appendices intended to minimize the need to refer to other mathematical sources. This textbook facilitates students' ability to apply fundamental principles and concepts in classical thermodynamics to solve challenging problems relevant to industry and everyday life. It also introduces the reader to the fundamentals of statistical mechanics, including understanding how the microscopic properties of atoms and molecules, and their associated intermolecular interactions, can be accounted for to calculate various average properties of macroscopic systems. The author emphasizes application of the fundamental principles outlined above to the calculation of a variety of thermodynamic properties, to the estimation of conversion efficiencies for work production by heat interactions, and to the solution of practical thermodynamic problems related to the behavior of non-ideal pure fluids and fluid mixtures, including phase equilibria and chemical reaction equilibria. The book contains detailed solutions to many challenging sample problems in classical thermodynamics and statistical mechanics that will help the reader crystallize the material taught. Class-tested and perfected over 30 years of use by nine-time Best Teaching Award recipient Professor Daniel Blankschtein of the Department of Chemical Engineering at MIT, the book is ideal for students

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of Chemical and Mechanical Engineering, Chemistry, and Materials Science, who will benefit greatly from in-depth discussions and pedagogical explanations of key concepts. Distills critical concepts, methods, and applications from leading full-length textbooks, along with the author's own deep understanding of the material taught, into a concise yet rigorous graduate and advanced undergraduate text; Enriches the standard curriculum with succinct, problem-based learning strategies derived from the content of 50 lectures given over the years in the Department of Chemical Engineering at MIT; Reinforces concepts covered with detailed solutions to illuminating and challenging homework problems.

This book contains a modern selection of about 200 solved problems and examples arranged in a didactic way for hands-on experience with course work in a standard advanced undergraduate/first-year graduate class in thermodynamics and statistical physics. The principles of thermodynamics and equilibrium statistical physics are few and simple, but their application often proves more involved than it may seem at first sight. This book is a comprehensive complement to any textbook in the field, emphasizing the analogies between the different systems, and paves the way for an in-depth study of solid state physics, soft matter physics, and field theory.

This text provides balanced coverage of the basic concepts of thermodynamics and heat transfer. Together with the illustrations, student-friendly writing style, and accessible math, this is an ideal text for an introductory thermal science course for non-mechanical engineering majors.

Learn classical thermodynamics alongside statistical mechanics and how macroscopic and microscopic ideas interweave with this fresh approach to the subjects.

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This book provides a comprehensive exposition of the theory of equilibrium thermodynamics and statistical mechanics at a level suitable for well-prepared undergraduate students. The fundamental message of the book is that all results in equilibrium thermodynamics and statistical mechanics follow from a single unprovable axiom — namely, the principle of equal a priori probabilities — combined with elementary probability theory, elementary classical mechanics, and elementary quantum mechanics.

Engel and Reid's *Thermodynamics, Statistical Thermodynamics, and Kinetics* gives students a contemporary and accurate overview of physical chemistry while focusing on basic principles that unite the sub-disciplines of the field. The Third Edition continues to emphasize fundamental concepts and presents cutting-edge research developments that demonstrate the vibrancy of physical chemistry today.

A comprehensive introduction to this important subject, presenting the fundamentals of classical and statistical thermodynamics through carefully developed concepts which are supported by many examples and applications. * Each chapter includes numerous carefully worked out examples and problems * Takes a more applied approach rather than theoretical * Necessary mathematics is left simple * Accessible to those fairly new to the subject

Statistical mechanics is concerned with defining the thermodynamic properties of a macroscopic sample in terms of the properties of the microscopic systems of which it is composed. The previous book *Introduction to Statistical Mechanics* provided a clear, logical, and self-contained treatment of equilibrium statistical mechanics starting from Boltzmann's two statistical assumptions, and presented a wide variety of applications to diverse physical assemblies. An appendix provided an introduction to non-equilibrium statistical mechanics

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through the Boltzmann equation and its extensions. The coverage in that book was enhanced and extended through the inclusion of many accessible problems. The current book provides solutions to those problems. These texts assume only introductory courses in classical and quantum mechanics, as well as familiarity with multi-variable calculus and the essentials of complex analysis. Some knowledge of thermodynamics is also assumed, although the analysis starts with an appropriate review of that topic. The targeted audience is first-year graduate students and advanced undergraduates, in physics, chemistry, and the related physical sciences. The goal of these texts is to help the reader obtain a clear working knowledge of the very useful and powerful methods of equilibrium statistical mechanics and to enhance the understanding and appreciation of the more advanced texts.

Introductory Statistical Thermodynamics is a text for an introductory one-semester course in statistical thermodynamics for upper-level undergraduate and graduate students in physics and engineering. The book offers a high level of detail in derivations of all equations and results. This information is necessary for students to grasp difficult concepts in physics that are needed to move on to higher level courses. The text is elementary, self contained, and mathematically well-founded, containing a number of problems with detailed solutions to help students to grasp the more difficult theoretical concepts. Beginning chapters place an emphasis on quantum mechanics Includes problems with detailed solutions and a number of detailed theoretical derivations at the end of each chapter Provides a high level of detail in derivations of all equations and results

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