

Assignment On Ionic And Covalent Compounds D Colgur

Steve and Susan Zumdahl's texts focus on helping students build critical thinking skills through the process of becoming independent problem-solvers. They help students learn to think like a chemists so they can apply the problem solving process to all aspects of their lives. In CHEMISTRY: AN ATOMS FIRST APPROACH, the Zumdahls use a meaningful approach that begins with the atom and proceeds through the concept of molecules, structure, and bonding, to more complex materials and their properties. Because this approach differs from what most students have experienced in high school courses, it encourages them to focus on conceptual learning early in the course, rather than relying on memorization and a plug and chug method of problem solving that even the best students can fall back on when confronted with familiar material. The atoms first organization provides an opportunity for students to use the tools of critical thinkers: to ask questions, to apply rules and models and to evaluate outcomes. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

This book includes 49 chapters presented as plenary , invited lectures and posters at the conference. Six plenary lectures have published in an issue of Pure and Applied Chemistry, Vol. 79, No. 12, 2007; the titles of these presentations are given as an Annex at the end of the book. I thank all contributors for the preparation of their presentations. It is sad to report that Professor Hitoshi Ohtaki, one of the founders of the Eurasia conferences and contributors passed away on November 5, 2006. Professor Ohtaki enthusiastically promoted international cooperation and took it upon himself to p- lictize Japanese science to the wider world. His contribution in this book will serve as a memorable contribution to that goal. He will be missed by all of us. This book is dedicated to his memory. Professor Dr . Bilge S ? ener Editor Memorial Tribute to Professor Dr. Hitoshi Ohtaki Curriculum Vitae of Hitoshi Ohtaki Date of Birth September 16, 1932 Place of Birth T ok yo, Japan Date of Decease November 5, 2006 (at the age of 74) Addr ess 3-9-406 Namiki-2-chome, Kanazawa-ku, Yokohama, Japan Institution Chair Professor of The Research Organization of Science and Engineering, Ritsumeikan University Guest Professor of Yokohama City University Education Bachelor of Science, Nagoya University, 1955 Master of Science, Nagoya University, 1957 Doctor of Science, Nagoya University, 1961 ix x Memorial Tribute to Professor Dr.

Class-tested and thoughtfully designed for student engagement, Principles of Organic Chemistry provides the tools and foundations needed by students in a short course or one-semester class on the subject. This book does not dilute the material or rely on rote memorization. Rather, it focuses on the underlying principles in order to make accessible the science that underpins so much of our day-to-day lives, as well as present further study and practice in medical and

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scientific fields. This book provides context and structure for learning the fundamental principles of organic chemistry, enabling the reader to proceed from simple to complex examples in a systematic and logical way. Utilizing clear and consistently colored figures, Principles of Organic Chemistry begins by exploring the step-by-step processes (or mechanisms) by which reactions occur to create molecular structures. It then describes some of the many ways these reactions make new compounds, examined by functional groups and corresponding common reaction mechanisms. Throughout, this book includes biochemical and pharmaceutical examples with varying degrees of difficulty, with worked answers and without, as well as advanced topics in later chapters for optional coverage. Incorporates valuable and engaging applications of the content to biological and industrial uses Includes a wealth of useful figures and problems to support reader comprehension and study Provides a high quality chapter on stereochemistry as well as advanced topics such as synthetic polymers and spectroscopy for class customization

This Squid Ink Classic includes the full text of the work plus MLA style citations for scholarly secondary sources, peer-reviewed journal articles and critical essays for when your teacher requires extra resources in MLA format for your research paper.

Life is produced by the interplay of water and biomolecules. This book deals with the physicochemical aspects of such life phenomena produced by water and biomolecules, and addresses topics including "Protein Dynamics and Functions", "Protein and DNA Folding", and "Protein Amyloidosis". All sections have been written by internationally recognized front-line researchers. The idea for this book was born at the 5th International Symposium "Water and Biomolecules", held in Nara city, Japan, in 2008.

This book deals with the electron density distribution in molecules and solids as obtained experimentally by X-ray diffraction. It is a comprehensive treatment of the methods involved, and the interpretation of the experimental results in terms of chemical bonding and intermolecular interactions. Inorganic and organic solids, as well as metals, are covered in the chapters dealing with specific systems. As a whole, this monograph is especially appealing because of its broad interface with numerous disciplines. Accurate X-ray diffraction intensities contain fundamental information on the charge distribution in crystals, which can be compared directly with theoretical results, and used to derive other physical properties, such as electrostatic moments, the electrostatic potential and lattice energies, which are accessible by spectroscopic and thermodynamic measurements. Consequently, the work will be of great interest to a broad range of crystallographers and physical scientists.

This book covers the synthesis, reactions, and properties of elements and inorganic compounds for courses in descriptive inorganic chemistry. It is suitable for the one-semester (ACS-recommended) course or as a supplement in general chemistry courses. Ideal for major and non-majors, the book incorporates rich

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graphs and diagrams to enhance the content and maximize learning. Includes expanded coverage of chemical bonding and enhanced treatment of Buckminster Fullerenes Incorporates new industrial applications matched to key topics in the text

Essentials of Coordination Chemistry: A Simplified Approach with 3D Visuals provides an accessible overview of this key, foundational topic in inorganic chemistry. Thoroughly illustrated within the book and supplemented by online 3D images and videos in full color, this valuable resource covers basic fundamentals before exploring more advanced topics of interest. The work begins with an introduction to the structure, properties, and syntheses of ligands with metal centers, before discussing the variety of isomerism exhibited by coordination compounds, such as structural, geometrical and optical isomerism. As thermodynamics and kinetics provide a gateway to synthesis and reactivity of coordination compounds, the book then describes the determination of stability constants and composition of complexes. Building upon those principles, the resource then explains a wide variety of nucleophilic substitution reactions exhibited by both octahedral and square planar complexes. Finally, the book discusses metal carbonyls and nitrosyls, special classes of compounds that can stabilize zero or even negative formal oxidation states of metal ions. Highlighting preparations, properties, and structures, the text explores the unique type of Metal-Ligand bonding which enable many interesting applications of these compounds. Thoughtfully organized for academic use, Essentials of Coordination Chemistry: A Simplified Approach with 3D Visuals encourages interactive learning. Advanced undergraduate and graduate students, as well as researchers requiring a full overview and visual understanding of coordination chemistry, will find this book invaluable. Includes valuable visual content through 3D images and videos in full color, available online Provides a valuable introduction to the study of organic and inorganic ligands with metal centers Discusses advanced topics including metal carbonyls and nitrosyls

Principles of Biology Biology 211, 212, and 213

The study of fire debris analysis is vital to the function of all fire investigations, and, as such, Fire Debris Analysis is an essential resource for fire investigators. The present methods of analysis include the use of gas chromatography and gas chromatography-mass spectrometry, techniques which are well established and used by crime laboratories throughout the world. However, despite their universality, this is the first comprehensive resource that addresses their application to fire debris analysis. Fire Debris Analysis covers topics such as the physics and chemistry of fire and liquid fuels, the interpretation of data obtained from fire debris, and the future of the subject. Its cutting-edge material and experienced author team distinguishes this book as a quality reference that should be on the shelves of all crime laboratories. Serves as a comprehensive guide to the science of fire debris analysis Presents both basic and advanced concepts in an easily readable, logical sequence Includes a full-color insert with figures that illustrate key concepts discussed in the text

This profusely illustrated book, by a world-renowned chemist and award-winning chemistry teacher, provides science students with an introduction to atomic and molecular structure and bonding. (This is a reprint of a book first published by Benjamin/Cummings, 1973.)

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A practical introduction to ionic compounds for both mineralogists and chemists, this book bridges the two disciplines. It explains the fundamental principles of the structure and bonding in minerals, and emphasizes the relationship of structure at the atomic level to the symmetry and properties of crystals. This is a great reference for those interested in the chemical and crystallographic properties of minerals.

A unique overview of the different kinds of chemical bonds that can be found in the periodic table, from the main-group elements to transition elements, lanthanides and actinides. It takes into account the many developments that have taken place in the field over the past few decades due to the rapid advances in quantum chemical models and faster computers. This is the perfect complement to "Chemical Bonding - Fundamentals and Models" by the same editors, who are two of the top scientists working on this topic, each with extensive experience and important connections within the community.

Organic Chemistry Study Guide: Key Concepts, Problems, and Solutions features hundreds of problems from the companion book, Organic Chemistry, and includes solutions for every problem. Key concept summaries reinforce critical material from the primary book and enhance mastery of this complex subject. Organic chemistry is a constantly evolving field that has great relevance for all scientists, not just chemists. For chemical engineers, understanding the properties of organic molecules and how reactions occur is critically important to understanding the processes in an industrial plant. For biologists and health professionals, it is essential because nearly all of biochemistry springs from organic chemistry. Additionally, all scientists can benefit from improved critical thinking and problem-solving skills that are developed from the study of organic chemistry. Organic chemistry, like any "skill", is best learned by doing. It is difficult to learn by rote memorization, and true understanding comes only from concentrated reading, and working as many problems as possible. In fact, problem sets are the best way to ensure that concepts are not only well understood, but can also be applied to real-world problems in the work place. Helps readers learn to categorize, analyze, and solve organic chemistry problems at all levels of difficulty Hundreds of fully-worked practice problems, all with solutions Key concept summaries for every chapter reinforces core content from the companion book

Designed for students in Nebo School District, this text covers the Utah State Core Curriculum for chemistry with few additional topics.

Conceptual Chemistry Volume I For Class XI

This book is ideal for use in a one-semester introductory course in physical chemistry for students of life sciences. The author's aim is to emphasize the understanding of physical concepts rather than focus on precise mathematical development or on actual experimental details. Subsequently, only basic skills of differential and integral calculus are required for understanding the equations. The end-of-chapter problems have both physiochemical and biological applications.

Chemical Modelling covers a wide range of disciplines and this Specialist Periodical Report is the first stop for any materials scientist, biochemist, chemist or molecular physicist wishing to acquaint themselves with major developments and current opinion in the applications and theory of chemical modelling. The topics covered are wide ranging with authors writing on clusters to modelling nanotubes and dynamics. Containing both comprehensive and critical reviews, this volume is an essential resource and convenient reference for any research group active in the field or chemical sciences library.

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This reference on current VB theory and applications presents a practical system that can be applied to a variety of chemical problems in a uniform manner. After explaining basic VB theory, it discusses VB applications to bonding problems, aromaticity and antiaromaticity, the dioxygen molecule, polyradicals, excited states, organic reactions, inorganic/organometallic reactions, photochemical reactions, and catalytic reactions. With a guide for performing VB calculations, exercises and answers, and numerous solved problems, this is the premier reference for practitioners and upper-level students.

There are more than 20 million chemicals in the literature, with new materials being synthesized each week. Most of these molecules are stable, and the 3-dimensional arrangement of the atoms in the molecules, in the various solids may be determined by routine x-ray crystallography. When this is done, it is found that this vast range of molecules, with varying sizes and shapes can be accommodated by only a handful of solid structures. This limited number of architectures for the packing of molecules of all shapes and sizes, to maximize attractive intermolecular forces and minimizing repulsive intermolecular forces, allows us to develop simple models of what holds the molecules together in the solid. In this volume we look at the origin of the molecular architecture of crystals; a topic that is becoming increasingly important and is often termed, crystal engineering. Such studies are a means of predicting crystal structures, and of designing crystals with particular properties by manipulating the structure and interaction of large molecules. That is, creating new crystal architectures with desired physical characteristics in which the molecules pack together in particular architectures; a subject of particular interest to the pharmaceutical industry. ORGANIC CHEMISTRY is a student-friendly, cutting edge introduction for chemistry, health, and the biological sciences majors. In the Eighth Edition, award-winning authors build on unified mechanistic themes, focused problem-solving, applied pharmaceutical problems and biological examples. Stepwise reaction mechanisms emphasize similarities among mechanisms using four traits: breaking a bond, making a new bond, adding a proton, and taking a proton away. Pull-out organic chemistry reaction roadmaps designed stepwise by chapter help students devise their own reaction pathways. Additional features designed to ensure student success include in-margin highlighted integral concepts, new end-of-chapter study guides, and worked examples. This edition also includes brand new author-created videos. Emphasizing "how-to" skills, this edition is packed with challenging synthesis problems, medicinal chemistry problems, and unique roadmap problems. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

Fluorinated Materials for Energy Conversion offers advanced information on the application of fluorine chemistry to energy conversion materials for lithium batteries, fuel cells, solar cells and so on. Fluorine compounds and fluorination techniques have recently gained important roles in improving the electrochemical

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characteristics of such energy production devices. The book therefore focuses on new batteries with high performance, the improvements of cell performance and the improvement of electrode and cell characteristics. The authors present new information on the effect of fluorine and how to make use of fluorination techniques and fluorine compounds. With emphasis on recent developments, this book is suitable for students, researchers and engineers working in chemistry, materials science and electrical engineering. Contains practical information, supported by examples Provides an update on recent developments in the field Written by specialists working in fluorine chemistry, electrochemistry, polymer and solid state chemistry

Strategies to achieve winning results in the inclusive secondary classroom! Backed by the author's three decades of experience, this reader-friendly guidebook provides teachers with a practical approach for creating a successful inclusive secondary classroom. Toby J. Karten helps teachers use a variety of strategies, including differentiated instruction, universal design for learning, brain-based learning, RTI, and evidence-based practice. With helpful forms, activities, graphic organizers, and quotations throughout, this resource: Outlines the theoretical background for creating an inclusive classroom environment Describes the psychosocial, cognitive, physical, and moral development of adolescents and how they affect teaching practice Provides research-based practices to maximize and honor learners' potentials and strengths

A book on Conceptual Chemistry

Concepts of Biology is designed for the single-semester introduction to biology course for non-science majors, which for many students is their only college-level science course. As such, this course represents an important opportunity for students to develop the necessary knowledge, tools, and skills to make informed decisions as they continue with their lives. Rather than being mired down with facts and vocabulary, the typical non-science major student needs information presented in a way that is easy to read and understand. Even more importantly, the content should be meaningful. Students do much better when they understand why biology is relevant to their everyday lives. For these reasons, Concepts of Biology is grounded on an evolutionary basis and includes exciting features that highlight careers in the biological sciences and everyday applications of the concepts at hand. We also strive to show the interconnectedness of topics within this extremely broad discipline. In order to meet the needs of today's instructors and students, we maintain the overall organization and coverage found in most syllabi for this course. A strength of Concepts of Biology is that instructors can customize the book, adapting it to the approach that works best in their classroom. Concepts of Biology also includes an innovative art program that incorporates critical thinking and clicker questions to help students understand--and apply--key concepts.

University Physics is designed for the two- or three-semester calculus-based physics course. The text has been developed to meet the scope and sequence of

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most university physics courses and provides a foundation for a career in mathematics, science, or engineering. The book provides an important opportunity for students to learn the core concepts of physics and understand how those concepts apply to their lives and to the world around them. Due to the comprehensive nature of the material, we are offering the book in three volumes for flexibility and efficiency. Coverage and Scope Our University Physics textbook adheres to the scope and sequence of most two- and three-semester physics courses nationwide. We have worked to make physics interesting and accessible to students while maintaining the mathematical rigor inherent in the subject. With this objective in mind, the content of this textbook has been developed and arranged to provide a logical progression from fundamental to more advanced concepts, building upon what students have already learned and emphasizing connections between topics and between theory and applications. The goal of each section is to enable students not just to recognize concepts, but to work with them in ways that will be useful in later courses and future careers. The organization and pedagogical features were developed and vetted with feedback from science educators dedicated to the project.

VOLUME III Unit 1: Optics
Chapter 1: The Nature of Light Chapter 2: Geometric Optics and Image Formation Chapter 3: Interference Chapter 4: Diffraction Unit 2: Modern Physics
Chapter 5: Relativity Chapter 6: Photons and Matter Waves Chapter 7: Quantum Mechanics Chapter 8: Atomic Structure Chapter 9: Condensed Matter Physics Chapter 10: Nuclear Physics Chapter 11: Particle Physics and Cosmology

Advances in Organometallic Chemistry

This volume, of a two volume set on ionic liquids, focuses on the applications of ionic liquids in a growing range of areas. Throughout the 1990s, it seemed that most of the attention in the area of ionic liquids applications was directed toward their use as solvents for organic and transition-metal-catalyzed reactions. Certainly, this interest continues on to the present date, but the most innovative uses of ionic liquids span a much more diverse field than just synthesis. Some of the main topics of coverage include the application of RTILs in various electronic applications (batteries, capacitors, and light-emitting materials), polymers (synthesis and functionalization), nanomaterials (synthesis and stabilization), and separations. More unusual applications can be noted in the fields of biomass utilization, spectroscopy, optics, lubricants, fuels, and refrigerants. It is hoped that the diversity of this volume will serve as an inspiration for even further advances in the use of RTILs.

Emphasises on contemporary applications and an intuitive problem-solving approach that helps students discover the exciting potential of chemical science. This book incorporates fresh applications from the three major areas of modern research: materials, environmental chemistry, and biological science.

Electrons, Atoms, and Molecules in Inorganic Chemistry: A Worked Examples Approach builds from fundamental units into molecules, to provide the reader with a full understanding of inorganic chemistry concepts through worked

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examples and full color illustrations. The book uniquely discusses failures as well as research success stories. Worked problems include a variety of types of chemical and physical data, illustrating the interdependence of issues. This text contains a bibliography providing access to important review articles and papers of relevance, as well as summaries of leading articles and reviews at the end of each chapter so interested readers can readily consult the original literature. Suitable as a professional reference for researchers in a variety of fields, as well as course use and self-study. The book offers valuable information to fill an important gap in the field. Incorporates questions and answers to assist readers in understanding a variety of problem types Includes detailed explanations and developed practical approaches for solving real chemical problems Includes a range of example levels, from classic and simple for basic concepts to complex questions for more sophisticated topics Covers the full range of topics in inorganic chemistry: electrons and wave-particle duality, electrons in atoms, chemical binding, molecular symmetry, theories of bonding, valence bond theory, VSEPR theory, orbital hybridization, molecular orbital theory, crystal field theory, ligand field theory, electronic spectroscopy, vibrational and rotational spectroscopy

Peter Atkins and Julio de Paula offer a fully integrated approach to the study of physical chemistry and biology.

Metal clusters are on the brink between molecules and nanoparticles in size. With molecular, nano-scale, metallic as well as non-metallic aspects, metal clusters are a growing, interdisciplinary field with numerous potential applications in chemistry, catalysis, materials and nanotechnology. This third volume in the series of hot topics from inorganic chemistry covers all recent developments in the field of metal clusters, with some 20 contributions providing an in-depth view. The result is a unique perspective, illustrating all facets of this interdisciplinary area:

- * Inter-electron Repulsion and Irregularities in the Chemistry of Transition Series
- * Stereochemical Activity of Lone Pairs in Heavier Main Group Element Compounds
- * How Close to Close Packing?
- * Forty-Five Years of Praseodymium Diodide
- * Centered Zirconium Clusters
- * Titanium Niobium Oxychlorides
- * Trinuclear Molybdenum and Tungsten Cluster Chalcogenides
- * Current State of (B,C,N)-Compounds of Calcium and Lanthanum
- * Ternary Phases of Lithium with Main-Group and Late-Transition Metals
- * Polar Intermetallics and Zintl Phases along the Zintl Border
- * Rare Earth Zintl Phases
- * Structure-Property Relationships in Intermetallics
- * Ternary and Quaternary Niobium Arsenide Zintl Phases
- * The Building Block Approach to Understanding Main-Group-Metal Complex Structures
- * Cation-Deficient Quaternary Thiospinels
- * A New Class of Hybrid Materials via Salt Inclusion Synthesis
- * Layered Perrhenate and Vanadate Hybrid Solids
- * Hydrogen Bonding in Metal Halides
- * Syntheses and Catalytic Properties of Titanium Nitride Nanoparticles
- * Solventless Thermolysis
- * New Potential Scintillation Materials in Borophosphate Systems.

With its didactical emphasis, this volume addresses a wide readership, such that both students and

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specialists will profit from the expert contributions.

Filling a gap in our systematic knowledge of gold, this monograph covers the fundamental aspects, while also considering new applications of gold compounds in catalysis, as nanoparticles, and their potential application as luminescent compounds. Written by an eminent team of authors from academia, the book analyzes the current status of gold chemistry, its special characteristics, oxidation states and main type of complexes, before going on to look at the synthesis of supramolecular aggregates due to the formation of gold-gold, gold-metal interactions or other secondary bonds. Final sections deal with LEDs, solvoluminescent and electroluminescent materials, liquid crystals and catalysis. While of interest to advanced chemistry students, this book is also useful for researchers interested in the chemistry of gold and its applications, as well as those involved in metal-metal interactions, heteronuclear chemistry or in the optical properties of coordination compounds.

The Principles of Biology sequence (BI 211, 212 and 213) introduces biology as a scientific discipline for students planning to major in biology and other science disciplines. Laboratories and classroom activities introduce techniques used to study biological processes and provide opportunities for students to develop their ability to conduct research.

This is the first supplementary volume on silicon-nitrogen compounds. It begins with the system silicon-nitrogen and continues with the binary silicon nitrides, except for Si_3N_4 that will be the subject of a separate Handbook volume. A chapter describes anionic formula units of the type SixNr and includes a condensed description of a series of ternary silicon nitrides. A treatment follows of the molecular and polymeric silicon-nitrogen-hydrogen compounds in which hydrogen is bonded either to silicon, or to nitrogen, or to both. In other words, those compounds have been selected for this chapter in which one or more nitrogen atoms are bonded to silicon and also those compounds in which additional silicon and/or hydrogen have been bonded to the initial silicon. About 60% of this volume contains a description of predominantly molecular silicon nitrogen compounds that are substituted at the nitrogen with groups containing other elements. According to the Gmelin System the substituting groups may be bonded through sulfur, boron, carbon, or both sulfur and carbon, or both boron and carbon. As may be expected, the largest quantity of information exists for compounds that have organyl substituents on the nitrogen, including the class of silicon coordination compounds that fulfill the criteria. In particular in these chapters, sections and subsections have been formed grouping compounds of the same class together. Compounds containing silicon--carbon bonds any where in the molecule are not covered in this volume. General remarks have been given to introduce chapters and to link chapters and sections.

Immunosensing for Detection of Protein Biomarkers not only introduces the principles, methods, and classification of immunoassay, but also presents the latest achievements in areas such as electrochemical immunosensors, nanoprobe-based immunoassay, chemiluminescence immunoassay, electrochemiluminescent immunoassay, multianalyte immunoassay, optical imaging for immunoassay, signal amplification for immunoassay, and so on. In recent years, immunosensing and immunoassay methods have attracted considerable interest due to their applications in different fields, particularly clinical diagnosis. Although a large number of academic papers in immunosensing and immunoassay have been published in different journals recently, it is still a difficult and time-consuming task for researchers, especially those new to the area, to understand the principles, methods, and research progress of immunosensing. Based on the research experience of the authors and their research groups, this book offers readers with new research ideas to develop immunosensing methodology. As a monograph, it offers deeper and more complete coverage than review papers, which only report certain aspects of progress. Grounded in the research experience of Professor Ju's research group, the book focuses on immunosensing for detection of protein biomarkers, summarizing understanding, research, and practice on immunosensing

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methodology in detection of protein biomarkers. Presents the latest research and thinking on immunosensing for detection of protein biomarkers Offers current techniques, and looks to the development of new methodologies Offers the latest developments in various aspects of immunosensing, including electrochemiluminescent immunoassay, multianalyte immunoassay, optical imaging immunoassay, and signal amplification immunoassay Offers readers new ideas to research and develop immunosensing methodologies for the future

From New York Times bestselling author Sam Kean comes incredible stories of science, history, finance, mythology, the arts, medicine, and more, as told by the Periodic Table. Why did Gandhi hate iodine (I, 53)? How did radium (Ra, 88) nearly ruin Marie Curie's reputation? And why is gallium (Ga, 31) the go-to element for laboratory pranksters?* The Periodic Table is a crowning scientific achievement, but it's also a treasure trove of adventure, betrayal, and obsession. These fascinating tales follow every element on the table as they play out their parts in human history, and in the lives of the (frequently) mad scientists who discovered them. THE DISAPPEARING SPOON masterfully fuses science with the classic lore of invention, investigation, and discovery--from the Big Bang through the end of time. *Though solid at room temperature, gallium is a moldable metal that melts at 84 degrees Fahrenheit. A classic science prank is to mold gallium spoons, serve them with tea, and watch guests recoil as their utensils disappear.

Most people remember chemistry from their schooldays as largely incomprehensible, a subject that was fact-rich but understanding-poor, smelly, and so far removed from the real world of events and pleasures that there seemed little point, except for the most introverted, in coming to terms with its grubby concepts, spells, recipes, and rules. Peter Atkins wants to change all that. In this Very Short Introduction to Chemistry, he encourages us to look at chemistry anew, through a chemist's eyes, in order to understand its central concepts and to see how it contributes not only towards our material comfort, but also to human culture. Atkins shows how chemistry provides the infrastructure of our world, through the chemical industry, the fuels of heating, power generation, and transport, as well as the fabrics of our clothing and furnishings. By considering the remarkable achievements that chemistry has made, and examining its place between both physics and biology, Atkins presents a fascinating, clear, and rigorous exploration of the world of chemistry - its structure, core concepts, and exciting contributions to new cutting-edge technologies. ABOUT THE SERIES: The Very Short Introductions series from Oxford University Press contains hundreds of titles in almost every subject area. These pocket-sized books are the perfect way to get ahead in a new subject quickly. Our expert authors combine facts, analysis, perspective, new ideas, and enthusiasm to make interesting and challenging topics highly readable.

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